

**Science (086)**  
**Class X**  
**Sample Question Paper 2022-23**

**Max. Marks: 80**

**Time Allowed: 3 hours**

**General Instructions:**

- i. This question paper consists of 39 questions in 5 sections.
- ii. All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- iii. **Section A** consists of 20 objective type questions carrying 1 mark each.
- iv. **Section B** consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- v. **Section C** consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- vi. **Section D** consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii. **Section E** consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

**SECTION - A**

Select and write one most appropriate option out of the four options given for each of the questions 1 – 20

Q. No	Questions	Marks
1	<p>The change in colour of the moist litmus paper in the given set up is due to</p>  <p>                     i. presence of acid                      ii. presence of base                      iii. presence of <math>H^+(aq)</math> in the solution                      iv. presence of Litmus which acts as an indicator                      (a) i and ii                      (b) Only ii                      (c) Only iii                      (d) Only iv.                 </p>	1
2	<p>In the redox reaction</p> $MnO_2 + 4HCl \rightarrow MnCl_2 + 2H_2O + Cl_2$ <p>                     (a) <math>MnO_2</math> is reduced to <math>MnCl_2</math> &amp; <math>HCl</math> is oxidized to <math>H_2O</math>                      (b) <math>MnO_2</math> is reduced to <math>MnCl_2</math> &amp; <math>HCl</math> is oxidized to <math>Cl_2</math> </p>	1

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