

Polycom® SoundStation® IP Conference Phones

Product Reference Guide

The family of Polycom® SoundStation® IP conference phones delivers unbeatable clarity and performance for any size room—from executive offices to board rooms. Don't settle for a desk phone in your conference room—only conference phones can deliver the audio performance and 360° microphone pickup required for efficient and productive conference calls.

SoundStation IP 7000 – Astounding voice quality and clarity from the world's most advanced IP conference phone

HDvoice



- 22 kHz CD-quality Polycom HD Voice™ technology for life-like clarity and intelligibility
- 20-foot (6.1-meter) microphone pickup range, more with optional expansion mics
- Resists interference from mobile phones and other wireless devices
- Connect two units together for additional microphone pickup and speaker volume
- Large high resolution display and processing power for IP applications
- Integration with Polycom HDX™ high definition video conferencing systems
- Integrated Power over Ethernet (PoE) connectivity
- Applications Port for mobile phone and PC connectivity

SoundStation IP 7000 – PoE Only	2200-40000-001
SoundStation IP 7000 – with AC Power Kit	2230-40300-001*

SoundStation IP 6000 – Next-generation IP conference phone designed for small to midsize rooms

HDvoice



- 14 kHz Polycom HD Voice for remarkable clarity and intelligibility
- 12-foot (3.6-meter) microphone pickup range, more with optional expansion mics
- Resists interference from mobile phones and other wireless devices
- High resolution backlit display for vital call information and multi-language support
- Integrated Power over Ethernet (PoE) connectivity

SoundStation IP 6000 – PoE Only	2200-15600-001
SoundStation IP 6000 – with AC Power Kit	2200-15650-001*

SoundStation IP 5000 – Advanced IP conference phone designed for small conference rooms and executive offices

HDvoice



- 7 kHz Polycom HD Voice for remarkable clarity and intelligibility
- 7-foot (2.1-meter) microphone pickup range
- Resists interference from mobile phones and other wireless devices
- High resolution backlit display for vital call information and multi-language support
- Integrated Power over Ethernet (PoE) connectivity
- Small footprint is ideal for executive offices and small conference tables

SoundStation IP 5000 – PoE Only	2200-30900-005
SoundStation IP 5000 – AC Power Kit	2200-43240-001*
Accessory	

* North America part number for AC Power Kits, please refer to local channel partner for correct part numbers in other regions

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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lab report 15 nitration of bromobenzene brynn abbott amp katie muerer bea0006 uah ch 336 02 september 21 2022 abstract in this experiment the electrophilic aromatic substitution mechanism was used to obtain 4 nitrobromobenzene and 2 nitrobromobenzene recrystallization was used to purify 4 nitrobromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0ml of nitric acid and 4.0ml of sulfuric acid and will then react with 4.5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

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chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment

nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group