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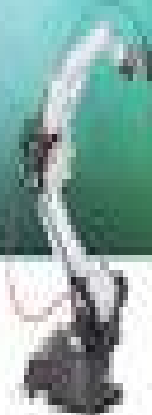
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Panasonic C Series Robots Manual

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 $80 \text{ g I}_2\text{O}_5 \times \frac{1 \text{ mol I}_2\text{O}_5}{1 \text{ mol I}_2} \times \frac{1 \text{ mol I}_2}{1 \text{ mol I}_2\text{O}_5} = 28 \text{ g CO} \times \frac{1 \text{ mol CO}}{1 \text{ mol CO}} = 1 \text{ mol CO}$

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 percent yield = $\frac{\text{actual yield}}{\text{theoretical yield}} \times 100$ label 3 7 3