

# AP Chemistry: A Sample Formal Laboratory Report

*This paper is designed to help you prepare a chemistry lab report. Keep it in your chemistry notebook. All chemistry lab reports must be written neatly and well organized to receive full credit. Lab reports may be written or typed. It is highly recommended that you use graph or engineering bond paper for written reports.*

## Laboratory #7: Quantitative Determination of an Empirical Formula

**I. Hypothesis:** If nitric acid is poured onto tin, a tin oxide will be produced. If we know the initial mass of the tin metal and the mass of the final product, we can determine the empirical formula of the tin oxide product. There should be a whole number ratio between oxygen and tin. *(The hypothesis explains what is to be tested and will be written after reading the entire laboratory worksheet.)*

**II. Equipment:** *(Non-chemical equipment used in the experiment.)*  
evaporating dish  
watch glass  
stirring rod  
burner with ring stand, ring and wire gauze  
forceps  
beaker  
balance

**III. Reagents:** *(A listing of chemicals used in the experiment with their amounts and any warnings.)*  
tin metal (granulated) ~2 g.  
5 cm<sup>3</sup> (mL) nitric acid (HNO<sub>3</sub>) **caution! severe burns**

**IV. Procedure:** Each step of the procedure must be written here. You may paraphrase and shorten the explanations, but the reader must be able to perform the experiment from these instructions. *(The procedure must be read carefully before the lab begins. Drawings of the experimental setup may be included here. The teacher may make changes to the procedure; make sure that you write any changes down!)*

**V. Data:** *(If the laboratory report is handwritten use a ruler to draw data tables and graphs! Always include units with all data entries.)*

|   | Procedure                        | Trial 1   | Trial 2 |
|---|----------------------------------|-----------|---------|
| a | mass of dish, and watch glass    | 74.14 g   |         |
| b | mass of dish, glass and tin      | 76.20 g   |         |
| c | mass of tin =b-a                 | 2.06 g    |         |
| d | moles of tin                     | .0173 mol |         |
| e | mass of dish, glass, and product | 76.76 g   |         |
| f | mass of oxygen =e-b              | .56 g     |         |
| g | moles of oxygen                  | .0350 mol |         |
| h | mole ratio                       | 2.02 : 1  |         |
| i | accepted ratio                   | 2 : 1     |         |
| j | % error                          | 1.00 %    |         |

# Sample Chemistry Laboratory Report

**Siddappa N.Byrareddy**



## **Sample Chemistry Laboratory Report:**

## Whispering the Techniques of Language: An Psychological Quest through **Sample Chemistry Laboratory Report**

In a digitally-driven world wherever screens reign great and immediate communication drowns out the subtleties of language, the profound techniques and mental subtleties concealed within words usually go unheard. However, situated within the pages of **Sample Chemistry Laboratory Report** a fascinating literary value pulsing with fresh emotions, lies an extraordinary quest waiting to be undertaken. Composed by a talented wordsmith, this enchanting opus invites viewers on an introspective trip, gently unraveling the veiled truths and profound affect resonating within ab muscles fabric of every word. Within the emotional depths of this touching evaluation, we can embark upon a honest exploration of the book is primary themes, dissect their fascinating publishing fashion, and succumb to the strong resonance it evokes heavy within the recesses of readers hearts.

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