

Chemical Equilibrium—Reversible Reactions

Refer to equilibrium equations in the discussion when answering these questions.

A. Saturated Sodium Chloride

1. What is the evidence for a shift in equilibrium? _____
2. Which ion caused the equilibrium to shift? _____
3. In which direction did the equilibrium shift? _____
4. If solid sodium hydroxide were added to neutralize the hydrochloric acid, would this reverse the reaction and cause the precipitated sodium chloride to redissolve? Explain.

B. Saturated Ammonium Chloride

1. What is the evidence for a shift in equilibrium? _____
2. In which direction did the equilibrium shift? _____
3. Which ion caused the equilibrium to shift? _____

C. Iron(III) Chloride plus Potassium Thiocyanate

1. What is the evidence for a shift in equilibrium when iron(III) chloride is added to the stock solution?
2. What is the evidence for a shift in equilibrium when potassium thiocyanate is added to the stock solution?

Report For Experiment 24 Reversible Reactions

D Kirk



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Foundations of College Chemistry, Laboratory Morris Hein, Susan Arena, 2010-08-09 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis *Scientific and Technical Aerospace Reports* ,1995 *Foundations of Chemistry in the Laboratory* Morris Hein, Leo R. Best, Robert L. Miner, 1990 **Report of the Agricultural Experiment Station of the University of California** ,1924 **Report of the Agricultural Experiment Station of the University of California** California Agricultural Experiment Station, 1924 U.S. Government Research & Development Reports ,1970

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