

REPORT FOR EXPERIMENT 9

Single Displacement Reactions

NOTE: In this experiment, copper ions will have 2+ charge.

NAME Tate
LAB PARTNER(S) _____
DATE 11/1/20

Evidence of Reaction

Describe any evidence of reaction; if no reaction was observed, write "None".

Equation (to be completed)
Write "No reaction", if no reaction was observed.

1. Changed color decomposing



2. change in color sparkly



3. none



4. none



5. bubbling/dissolving



6. none



QUESTIONS AND PROBLEMS

1. Complete the following table by writing the symbols of the two elements whose reactivities are being compared in each test:

	Tube Number					
	1	2	3	4	5	6
Greater activity	<u>Cu</u>	<u>Zn</u>	<u>Mg</u>	<u>H</u>	<u>Zn</u>	<u>Pb</u>
Lesser activity	<u>Ag</u>	<u>Pb</u>	<u>Zn</u>	<u>Cu</u>	<u>H</u>	<u>Cu</u>

2. Arrange Pb, Mg, and Zn in order of their activities, listing the most active first.

(1) Mg

(2) Zn

(3) Pb

Report For Experiment 1single Displacement Reactions

Arturo Cuomo



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Table of Contents Report For Experiment 1single Displacement Reactions

1. Understanding the eBook Report For Experiment 1single Displacement Reactions
 - The Rise of Digital Reading Report For Experiment 1single Displacement Reactions
 - Advantages of eBooks Over Traditional Books
2. Identifying Report For Experiment 1single Displacement Reactions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Report For Experiment 1single Displacement Reactions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Report For Experiment 1single Displacement Reactions
 - Personalized Recommendations
 - Report For Experiment 1single Displacement Reactions User Reviews and Ratings
 - Report For Experiment 1single Displacement Reactions and Bestseller Lists
5. Accessing Report For Experiment 1single Displacement Reactions Free and Paid eBooks
 - Report For Experiment 1single Displacement Reactions Public Domain eBooks
 - Report For Experiment 1single Displacement Reactions eBook Subscription Services
 - Report For Experiment 1single Displacement Reactions Budget-Friendly Options

6. Navigating Report For Experiment 1single Displacement Reactions eBook Formats
 - ePub, PDF, MOBI, and More
 - Report For Experiment 1single Displacement Reactions Compatibility with Devices
 - Report For Experiment 1single Displacement Reactions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Report For Experiment 1single Displacement Reactions
 - Highlighting and Note-Taking Report For Experiment 1single Displacement Reactions
 - Interactive Elements Report For Experiment 1single Displacement Reactions
8. Staying Engaged with Report For Experiment 1single Displacement Reactions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Report For Experiment 1single Displacement Reactions
9. Balancing eBooks and Physical Books Report For Experiment 1single Displacement Reactions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Report For Experiment 1single Displacement Reactions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Report For Experiment 1single Displacement Reactions
 - Setting Reading Goals Report For Experiment 1single Displacement Reactions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Report For Experiment 1single Displacement Reactions
 - Fact-Checking eBook Content of Report For Experiment 1single Displacement Reactions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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