

Answer Key

Guided Reading 6.1

Observing Chemical Change Guided Reading and Study Use Target Reading Skills Sample questions and answers: Question What are physical properties of matter? Answer Physical properties are characteristics that can be observed without changing one substance into another. Question What is the evidence for chemical reactions? Answer The evidence for chemical reactions is the formation of new substances and changes in energy.

1. Matter is anything that has mass and takes up space.
2. chemistry
3. a. Physical b. Only by changing one substance into another
4. False
5. a, b, d
6. chemical reaction, or chemical change
7. They break and form new bonds, producing new substances.
8. Formation of new substances and changes in energy
9. The change in color is an indication of a change in properties, and a change in properties is a sign that a chemical reaction occurred.
10. precipitate
11. It might indicate a chemical reaction, because the bubbles are gas, and the presence of a gas indicates a new substance may have been formed.
12. True
13. A change in temperature reflects a change in energy.
14. False
15. a. Energy is absorbed b. Exothermic

6.2

Describing Chemical Reactions Guided Reading and Study Use Target Reading Skills a. A chemical equation is a short, easy way to show a chemical reaction. b. A substance you have at the beginning of a reaction is a reactant. c. A new substance produced in a reaction is a product. d. Conservation of mass means that during a chemical reaction, matter is not created or destroyed. e. In an open system, matter can enter from or escape to the surroundings. f. In a closed system, matter is not allowed to enter or leave. g. A coefficient is a number in a chemical equation telling you how many atoms or molecules of a

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Michael Hoelscher



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