

Study Guide

Heat of Fusion & Phase Change

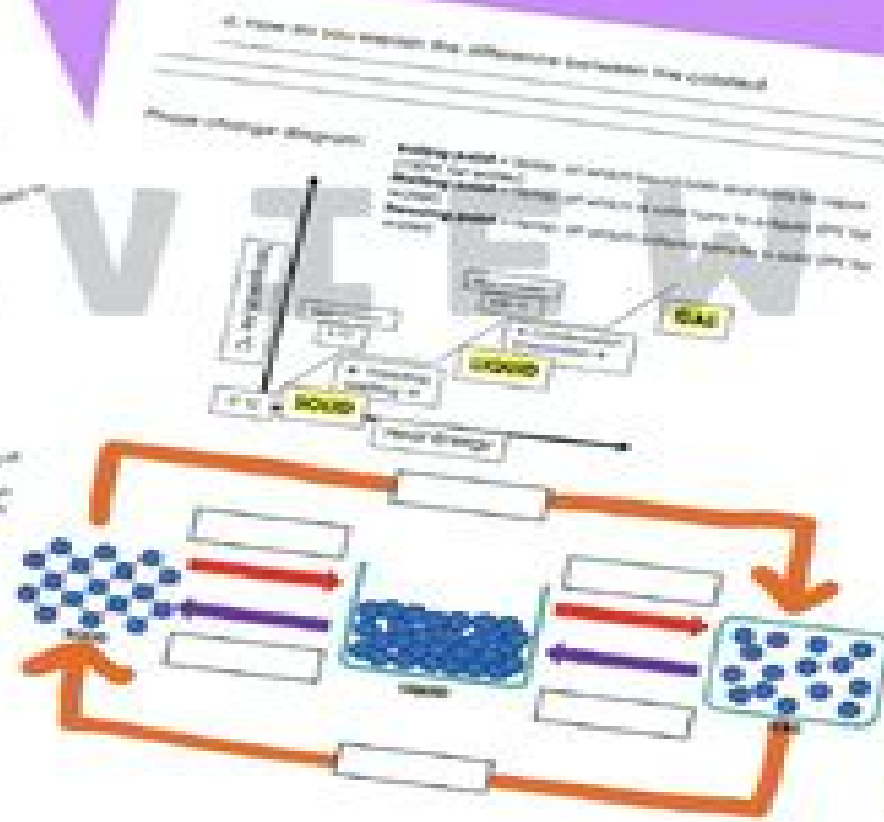
Name: _____ Date of Quiz: _____

Chemistry Investigations G & I

SAIDS GUIDE-

Heat of Fusion & Phase Change

- Heat of fusion is the energy required to change a substance from a solid to a liquid at its melting point.
- Heat of fusion is a constant for each substance.
1. Draw a graph of heat of fusion vs. temperature for water.
 2. Label the temperature of the melting point.
 3. Label the temperature of the boiling point.
 4. Label the temperature of the freezing point.
 5. Label the temperature of the condensation point.
 6. Label the temperature of the sublimation point.
 7. Label the temperature of the deposition point.
 8. Label the temperature of the solidification point.
 9. Label the temperature of the vaporization point.
 10. Label the temperature of the liquefaction point.
 11. Label the temperature of the solidification point.
 12. Label the temperature of the deposition point.
 13. Label the temperature of the sublimation point.
 14. Label the temperature of the liquefaction point.
 15. Label the temperature of the solidification point.
 16. Label the temperature of the deposition point.
 17. Label the temperature of the sublimation point.
 18. Label the temperature of the liquefaction point.
 19. Label the temperature of the solidification point.
 20. Label the temperature of the deposition point.



Phase Change Study Guide And Answers

M Lipman



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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