

Name: _____ Date: _____

pH and pOH

Calculation Problems:

For the following questions, show your calculations.

- Calculate the pH of a solution with a hydrogen ion concentration of $1 \times 10^{-4} \text{ M}$.
- A solution has a pH of 2. What is the concentration of H^+ ions in this solution?
- What is the pH of a solution with a hydroxide ion concentration of $1 \times 10^{-6} \text{ M}$?
- Calculate the pOH of a solution where the concentration of OH^- ions is $3.0 \times 10^{-3} \text{ M}$.
- A solution has a pOH of 6. What is the pH of this solution?

Ph And Poh Continued Problems With Answers

SB Merriam



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