

Periodic and Groups: Valence Electrons: Mastery

Periodic and Groups

Elements that have the same number of valence electrons have similar properties, so we put them in groups. The the periodic table groups + columns that look like this. They will see family groups 1A-18A. Hydrogen is in the same period as Helium.

Elements that have electrons in the outer shell have similar properties in the Periodic Table. Periods go left to right. Elements in the same vertical columns

Period	Group		Group										18A	
	1A	2A	3A	4A	5A	6A	7A	8A	9A	10A	11A	12A	13A	14A
1	H 1H	He 2He											Li 3Li	Be 4Be
2	B 5B	C 6C	N 7N	O 8O	F 9F	Ne 10Ne								Na 11Na
3	Al 13Al	Si 14Si	P 15P	S 16S	Cl 17Cl	Ar 18Ar								K 19K
4	Ca 20Ca	Sc 21Sc	Ti 22Ti	V 23V	Cr 24Cr	Mn 25Mn	Fe 26Fe	Co 27Co	Ni 28Ni	Cu 29Cu	Zn 30Zn	Ga 31Ga	Ge 32Ge	Br 35Br

Thinking like the game "Monopoly", answer that an element is in period 3 and group 1.

What element is in Group 1A and Period 2?

What group and period is Helium in?

What element is in Group 2, and Period 10A? Helium, Lithium, B, Group 13A is Helium, 13A. The element is oxygen.

What element is in Group 2A, and Period 2?

What element Helium?

Valence Electrons

Valence Electrons are the most important electrons in an atom. Each period contains has the same number of valence electrons. Valence electrons are the electrons that are involved in chemical bonding.

Hydrogen 1



1 valence electron

Lithium 3



3 valence electrons

Group 17



7 valence electrons

Group 16, tellurium 54A and has 6 valence electrons. How many valence electrons will tellurium have?

How many electrons?

Hydrogen and Helium are both in Group 1A-2Helium, 2 valence electrons.

Group 18A - Elements in group 18A have a full shell of electrons. Elements in group 18A and 18A are called noble gases. Helium, neon, argon, krypton, xenon, and radon are noble gases. "He, Ne, Ar, Kr, Xe, Rn" Group 18A has a full shell of valence electrons. Helium, 2 valence electrons. Argon, 8 valence electrons. Krypton, 18 valence electrons. Xenon, 18 valence electrons. Radon, 18 valence electrons.

Periods And Groups Valence Electrons Masses Answers

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A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the EduGorilla Prep Experts text.

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