

- ____ 7. A kilogram(s) contains how many grams?
- 0.1×10^3
 - 0.1×10^6
 - 0.1
 - 0.001
 - 0.1×10^{-3}
- ____ 8. Convert 0.0002 m to mm.
- 200 mm
 - $2 \times 10^{-5} \text{ mm}$
 - $2 \times 10^5 \text{ mm}$
 - 2000 mm
 - 2 mm
- ____ 9. A conversion factor that shows the relationship between grams and kilograms is
- $\frac{1000 \text{ g}}{1 \text{ kg}}$
 - $\frac{1 \text{ kg}}{1000 \text{ g}}$
 - $\frac{1000}{1}$
 - $\frac{1}{1000}$
 - $\frac{1 \text{ g}}{1000 \text{ kg}}$
- ____ 10. We generally report a measurement by recording all of the certain digits plus _____ uncertain digit(s).
- one
 - two
 - three
 - four
 - five
- ____ 11. In the measurement 0.003 L, which digit is the estimated digit?
- 0
 - the 0 immediately to the left of the 3
 - 3
 - the 0 further left of the decimal point
- ____ 12. How many significant figures are there in the number 0.00456789?
- 4
 - 5
 - 7
 - 8
 - 9

New Mexico End Of Course Chemistry Exam

Patrick Vollmar



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