#### Introduction

Take a few minutes to play with the sim (<a href="http://phet.colorado.edu/en/simulation/acid-base-solutions">http://phet.colorado.edu/en/simulation/acid-base-solutions</a>). Check out both the Introduction and Custom Solution tabs. Explore what factors affect pH. List the factors you found that affect pH.

#### Investigating Concentration Changes

- 2. a. Create a strong acid solution in the "Custom Solution" tab.
  - b. Draw bar graphs for Initial and Equilibrium concentrations.
    - <u>Hints</u>: No calculator needed try the 'Equilibrium Concentration' view. Don't forget to label your graphs!

| Initial Concentrations | Equilibrium Concentrations |
|------------------------|----------------------------|
|                        |                            |
|                        |                            |
|                        |                            |
|                        |                            |
|                        |                            |
|                        |                            |

- c. What equilibrium concentrations are affected by changing the initial concentration?
- 3. a. Create a weak acid solution in the "Custom Solution" tab.
  - b. Draw bar graphs for Initial and Equilibrium concentrations.
    - <u>Hints</u>: No calculator needed try the 'Equilibrium Concentration' view. Don't forget to label your graphs!

| Initial Concentrations | Equilibrium Concentrations |
|------------------------|----------------------------|
|                        |                            |
|                        |                            |
|                        |                            |
|                        |                            |
|                        |                            |
|                        |                            |

- c. What equilibrium concentrations are affected by changing the initial concentration?
- 4. Are your results for the strong and weak acid in questions 2 and 3 consistent with the definition of strong and weak acids?

# **Phet Charles Law Computer Activity Answers**

M Mark

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