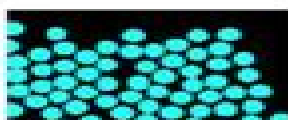


PhET Interactive Simulation: States of Matter

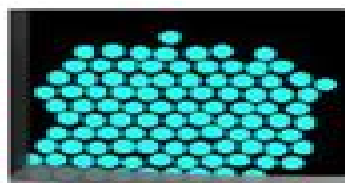
[Click Here for the States of Matter Simulation](#), practice then answer the questions below:
(Be sure to download the simulation before starting).

Exploration

1. Identify the following as either a solid, liquid, or a gas.



a) liquid



b) solid



c) gas

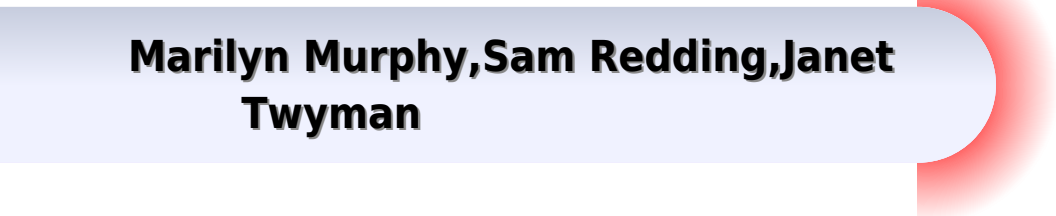
2. a) What happens to the atoms of a solid when the temperature increases?
b) decreases? *When the temp increases in a solid the atoms start to spread out, when the temp decreases the atoms get more and more compact.*
3. What happens to the atoms of a solid when the temperature decreases?
The atoms get more and more compact.
4. Complete the following table for the initial temperature of each substance in each state:

	Solid	Liquid	Gas
Neon	-295 c	-246 c	-217 c
Argon	-230 c	-187 c	-84 c
Oxygen	-246 c	-216 c	-160 c
Water	-127 c	13 c	156 c

5. Use the table above to compare any two (2) of the gases. How do their initial temperatures compare at each physical state?
When neon gets very cold it becomes a gas as you can see from the chart, but when water gets very hot it becomes a gas.
6. a) What happens to the pressure if the temperature is increased?
b) decreased? *When the temp is increased the pressure goes up for everyone of them except oxygen and when the temp is decreased the pressure goes down for everyone except oxygen. For oxygen when the temp is increased pressure goes down while decreasing it makes the pressure go up.*

Phet States Of Matter Lab Answers

**Marilyn Murphy, Sam Redding, Janet
Twyman**



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Handbook on Personalized Learning for States, Districts, and Schools Marilyn Murphy, Sam Redding, Janet Twyman, 2016-07-01 The recent passage of the Every Student Succeeds Act ESSA presents new opportunities and greater flexibility in efforts to personalize learning for all children The Handbook on Personalized Learning for States Districts and Schools provides insight and guidance on maximizing that new flexibility Produced by the Center on Innovations in Learning CIL one of seven national content centers funded by the U S Department of Education this volume suggests how teachers can enhance personalized learning by cultivating relationships with students and their families to better understand a child s learning and motivation Personalized learning also encourages the development of students metacognitive social and emotional competencies thereby fostering students self direction in their own education one aimed at mastery of knowledge and skills and readiness for career and college Chapters address topics across the landscape of personalized learning including co designing instruction and learning pathways with students variation in the time place and pace of learning including flipped and blended classrooms and using technology to manage and analyze the learning process The Handbook s chapters include Action Principles to guide states districts and schools in personalizing learning *Polymer Science & Technology*, 1971 Chemical Abstracts, 1991 Army, Navy, Air Force Journal and Register, 1938 *Properties of Matter: Three States of Matter Gr. 5-8* George Graybill, 2015-09-01 This is the chapter slice Three States of Matter from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives **Understanding the Properties of Matter** Michael de Podesta, 2002-04-25 Understanding the Properties of Matter 2nd Edition takes a unique phenomenological approach to the presentation of matter materials and solid state physics After an overview of basic ideas and a reminder of the importance of measurement the author considers in turn gases solids liquids and phase changes For each topic the focus is on what happens After a preliminary examination of data on the properties of matter the author raises then addresses a series of questions concerning the data It is only in answering these questions that he adopts the theoretical approach to the properties of matter This approach can reawaken in readers the fascination for the subject that inspired some of the greatest physicists of our age Examples and extensive exercises reinforce the concepts A supporting Web site furnishes for free download a plethora of additional materials including Supplementary chapters on the band theory of solids

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