

# 5.4

## THE PERIODIC TABLE: ORGANIZING THE ELEMENTS

### SECTION REVIEW

#### Objectives

- Describe the origin of the periodic table
- Identify the position of groups, periods, and the transition metals in the periodic table

#### Key Terms

- |                           |                           |               |
|---------------------------|---------------------------|---------------|
| • periodic table          | • metals                  | • nonmetals   |
| • periods                 | • alkali metals           | • halogens    |
| • periodic law            | • alkaline earth metals   | • noble gases |
| • group                   | • transition metals       | • metalloids  |
| • representative elements | • inner transition metals |               |

#### Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

- The periodic table organizes the elements into vertical 1 1. \_\_\_\_\_ and horizontal 2 2. \_\_\_\_\_ in order of increasing 3 3. \_\_\_\_\_. The table is constructed so that elements that have similar chemical properties are in the same 4 4. \_\_\_\_\_. The elements in Groups 1A through 7A are called the 5 5. \_\_\_\_\_. The 6 6. \_\_\_\_\_ make up Group 0. The elements in Groups 2A and 3A are interrupted in periods 4 and 5 by the 7 7. \_\_\_\_\_. and in the periods 6 and 7 by the 8 8. \_\_\_\_\_. The Group 1A elements are called the 9 9. \_\_\_\_\_, and the Group 2A elements are called the 10 10. \_\_\_\_\_. The nonmetals of 11 11. \_\_\_\_\_ are called the halogens. Elements with properties that are intermediate between those of metals and nonmetals are called 12 12. \_\_\_\_\_.

# Periodic Table Section Review Answer Key

**Hongru Du**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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## **Table of Contents Periodic Table Section Review Answer Key**

1. Understanding the eBook Periodic Table Section Review Answer Key
  - The Rise of Digital Reading Periodic Table Section Review Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Section Review Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Section Review Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Section Review Answer Key
  - Personalized Recommendations
  - Periodic Table Section Review Answer Key User Reviews and Ratings
  - Periodic Table Section Review Answer Key and Bestseller Lists
5. Accessing Periodic Table Section Review Answer Key Free and Paid eBooks
  - Periodic Table Section Review Answer Key Public Domain eBooks

- Periodic Table Section Review Answer Key eBook Subscription Services
- Periodic Table Section Review Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table Section Review Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table Section Review Answer Key Compatibility with Devices
  - Periodic Table Section Review Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table Section Review Answer Key
  - Highlighting and Note-Taking Periodic Table Section Review Answer Key
  - Interactive Elements Periodic Table Section Review Answer Key
- 8. Staying Engaged with Periodic Table Section Review Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Section Review Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Section Review Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Section Review Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Section Review Answer Key
  - Setting Reading Goals Periodic Table Section Review Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Section Review Answer Key
  - Fact-Checking eBook Content of Periodic Table Section Review Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks

---

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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