

# PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18		
1 H Hydrogen 1.008	2 He Helium 4.003											3 Li Lithium 6.941	4 Be Beryllium 9.012	5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180
3 Na Sodium 22.990	4 Mg Magnesium 24.305											5 Al Aluminum 26.982	6 Si Silicon 28.086	7 P Phosphorus 30.974	8 S Sulfur 32.065	9 Cl Chlorine 35.453	10 Ar Argon 39.948		
4 K Potassium 39.098	5 Ca Calcium 40.078	6 Sc Scandium 44.956	7 Ti Titanium 47.88	8 V Vanadium 50.942	9 Cr Chromium 52.004	10 Mn Manganese 54.938	11 Fe Iron 55.845	12 Co Cobalt 58.933	13 Ni Nickel 58.693	14 Cu Copper 63.546	15 Zn Zinc 65.38	16 Ga Gallium 69.723	17 Ge Germanium 72.630	18 As Arsenic 74.922	19 Se Selenium 78.96	20 Br Bromine 79.904	21 Kr Krypton 83.80		
5 Rb Rubidium 85.468	6 Sr Strontium 87.62	7 Y Yttrium 88.906	8 Zr Zirconium 91.224	9 Nb Niobium 92.906	10 Mo Molybdenum 95.94	11 Tc Technetium 98.906	12 Ru Ruthenium 101.07	13 Rh Rhodium 102.905	14 Pd Palladium 106.42	15 Ag Silver 107.868	16 Cd Cadmium 112.415	17 In Indium 114.818	18 Sn Tin 118.710	19 Sb Antimony 121.757	20 Te Tellurium 127.603	21 I Iodine 126.905	22 Xe Xenon 131.29		
6 Cs Cesium 132.905	7 Ba Barium 137.327	8 La Lanthanum 138.905	9 Ce Cerium 140.12	10 Pr Praseodymium 140.908	11 Nd Neodymium 144.24	12 Pm Promethium 144.913	13 Sm Samarium 150.36	14 Eu Europium 151.964	15 Gd Gadolinium 157.25	16 Tb Terbium 158.925	17 Dy Dysprosium 162.50	18 Ho Holmium 164.930	19 Er Erbium 167.259	20 Tm Thulium 168.934	21 Yb Ytterbium 173.054	22 Lu Lutetium 174.967			
7 Fr Francium 223	8 Ra Radium 226	9 Ac Actinium 227	10 Th Thorium 232.038	11 Pa Protactinium 231.036	12 U Uranium 238.029	13 Np Neptunium 237.048	14 Pu Plutonium 244.064	15 Am Americium 243.061	16 Cm Curium 247.070	17 Bk Berkelium 247.070	18 Cf Californium 251.083	19 Es Einsteinium 252.083	20 Fm Fermium 257.103	21 Md Mendelevium 258.103	22 No Nobelium 259.103	23 Lr Lawrencium 262.103			

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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57 La Lanthanum 138.905	58 Ce Cerium 140.12	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.259	69 Tm Thulium 168.934	70 Yb Ytterbium 173.054	71 Lu Lutetium 174.967
89 Ac Actinium 227	90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.083	99 Es Einsteinium 252.083	100 Fm Fermium 257.103	101 Md Mendelevium 258.103	102 No Nobelium 259.103	103 Lr Lawrencium 262.103

## Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol <sup>1</sup>H), has a single proton and zero neutrons.

## Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

## Lithium

Lithium (from Greek: λίθος, lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

# Periodic Table Basic Speedy Study Guides

**Yi-Tong Ma**



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