QUESTION 3

3. 1.Learners made a mini volcano in science laboratory by adding sodium bicarbonate to ethanoic acid. They added 100 ml of a 0, 2 mol.dm⁻³ ethanoic acid solution to 10 g of NaHCO₃ to start the reaction of the volcano.

The balanced equation for this reaction is:

- 3.1.1 Define the term limiting reagent. (2)
- 3.1.2 Determine the limiting reagent in this reaction. (6)
- 3.1.3 Calculate the mass of the other substance in excess. (3)
- 3.1.4 Calculate the volume of CO₂ produced at STP. (4)
- 3.2 .A solution of potassium hydroxide (KOH) is prepared by dissolving 3,36 g crystals of KOH in 250 cm³ of water.

Calculate the concentration of the potassium hydroxide solution.

(4)

3.3 .25 cm³ of a potassium hydroxide solution of concentration 0,25 mol.dm-3 completely neutralises a dilute solution of sulphuric acid (H₂SO₄) in a flask.

The unbalanced equation below represents the reaction that takes place:

$$H_2SO_4$$
 (aq) + KOH (aq) \rightarrow K₂ SO₄ (aq) + H₂O (ℓ)

- 3.3.1 Balance the above equation. (2)
- 3.3.2 Calculate the mass of sulphuric acid in the flask. (5)

Physical Science Grade 11 Controlled Test 2 September 2014

AW Rasmussen

Physical Science Grade 11 Controlled Test 2 September 2014:

This book delves into Physical Science Grade 11 Controlled Test 2 September 2014. Physical Science Grade 11 Controlled Test 2 September 2014 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Physical Science Grade 11 Controlled Test 2 September 2014, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Physical Science Grade 11 Controlled Test 2 September 2014
 - o Chapter 2: Essential Elements of Physical Science Grade 11 Controlled Test 2 September 2014
 - Chapter 3: Physical Science Grade 11 Controlled Test 2 September 2014 in Everyday Life
 - Chapter 4: Physical Science Grade 11 Controlled Test 2 September 2014 in Specific Contexts
 - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of Physical Science Grade 11 Controlled Test 2 September 2014. This chapter will elucidate the essential principles that need to be understood to grasp Physical Science Grade 11 Controlled Test 2 September 2014 in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Physical Science Grade 11 Controlled Test 2 September 2014 in daily life. This chapter will showcase real-world examples of how Physical Science Grade 11 Controlled Test 2 September 2014 can be effectively utilized in everyday scenarios.
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