

Student Solutions Manual

to accompany

Ira N. Levine

PHYSICAL CHEMISTRY

Fifth Edition



Physical Chemistry Levine 5th Edition Solutions Manual

DJ Losen



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Student Solutions Manual to Accompany Physical Chemistry, Fifth Edition Ira N. Levine, 2002 Student Solutions Manual to accompany Physical Chemistry Ira Levine, 2008-07-11 Written by Ira Levine the Student Solutions Manual contains the worked out solutions to all of the problems in the text The purpose of the manual is help the student learn physical chemistry and as an incentive to work problems not as a way to avoid working problems The British Library General Catalogue of Printed Books 1976 to 1982 British Library, 1983 **Books in Print Supplement** , 2002 Scientific and Technical Books and Serials in Print , 1989 *Books in Print* , 1994 **Subject Guide to Books in Print** , 1993

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This book provides a unique approach to introduce undergraduate students to the concepts and methods of physical chemistry which are the foundational principles of Chemistry. The book introduces the student to the principles underlying the essential sub fields of quantum mechanics, atomic and molecular structure, atomic and molecular spectroscopy, statistical thermodynamics, classical thermodynamics, solutions and equilibria, electrochemistry, kinetics and reaction dynamics, macromolecules and organized molecular assemblies. Importantly, the book develops and applies these principles to supramolecular assemblies and supramolecular machines with many examples from biology and nanoscience. In this way, the book helps the student to see the frontier of modern physical chemistry developments. The book begins with a discussion of wave particle duality and proceeds systematically to more complex chemical systems in order to relate the story of physical chemistry in an intellectually coherent manner. The topics are organized to correspond with those typically given in each of a two course semester sequence. The first 13 chapters present quantum mechanics and spectroscopy to describe and predict the structure of matter, atoms, molecules and solids. Chapters 14 to 29 present statistical thermodynamics and kinetics and apply their principles to understanding equilibria, chemical transformations, macromolecular properties and supramolecular machines. Each chapter of the book begins with a simplified view of a topic and evolves to more rigorous description in order to provide the student and instructor flexibility to choose the level of rigor and detail that suits them best. The textbook treats important new directions in physical chemistry research, including chapters on macromolecules, principles of interfaces and films for organizing matter and supramolecular machines, as well as including discussions of modern nanoscience, spectroscopy and reaction dynamics throughout the text.

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