

Chapter 6: Chemical Bonds Review Answer Key

- What is the most stable group on the Periodic Table? Why is this group stable?
Noble Gases (Group 18) because they have 8 valence electrons which means their outer energy level is full.
- Name the most reactive group(s) on the Periodic Table? Why are they reactive?
Alkali Metals (Group 1) & Halogens (Group 17): Alkali Metals want to lose one valence electron in order to become stable and Halogens want to gain one valence electron to be stable.

- In the boxes below draw the Electron dot diagrams for:

Sodium	Calcium	Boron	Chlorine	Krypton
$\text{Na} \cdot$	$\text{Ca} \cdot$ $\cdot$	$\cdot \text{B} \cdot$ $\cdot$	$\begin{array}{c} \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \end{array}$	$\begin{array}{c} \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \\ \cdot \cdot \end{array}$

- Name the three (3) types of chemical bonds:
Ionic bonds Covalent bonds Polyatomic
- Define ionic bond:
The force that holds cations and anions together.
- Define covalent bond:
A chemical bond in which two atoms share a pair of valence electrons.
- Define polyatomic ion:
A covalently bonded group of atoms that has a positive or negative charge and acts as a unit.
- In an ionic bond, electrons are transferred.
- Who are the participants in an ionic bond? Metals and nonmetals
- An ion with a positive charge is a cation; an ion with a negative charge is an anion.
- In a covalent bond, electrons are shared.
- Who are the participants in a covalent bond? Two nonmetals
- Identify whether the following compounds are covalent (two nonmetals), ionic (a nonmetal and a metal), or polyatomic (both).
 - CaCl_2 IONIC (transferred)
 - H_2O COVALENT (shared)
 - BaSO_4 POLYATOMIC (both)
 - CH_4 COVALENT (shared)
 - LiBr IONIC (transferred)
 - NH_4Cl POLYATOMIC (both)
 - KI IONIC (transferred)

Physical Science Chapter 6 Chemical Bonds Wordwise Answers

Victor M. Corman



Physical Science Chapter 6 Chemical Bonds Wordwise Answers:

Physical Science Chapter 6 Chemical Bonds Wordwise Answers Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has be much more evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Physical Science Chapter 6 Chemical Bonds Wordwise Answers**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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