

SCIENCE
Part - I : Physical Sciences
(English version)

Parts A and B

[Max. Marks : 40]

Time : 1 hour 30 min.]

Instructions :

- (i) Read the question paper and understand every question thoroughly and write answers in given 1 hr. 30 min. time.
- (ii) 3 very short answer questions are there in **section-I**. Each question carries 2 marks. Answer all the questions. Write answer to each question in 3 to 4 sentences.
- (iii) 3 short answer questions are there in **section-II**. Each question carries 4 marks. Answer all the questions. Write answer to each question in 5 to 6 sentences.
- (iv) 3 essay type questions are there in **section-III**. Each question carries 6 marks. Answer any two questions. Write answer to each question in 8 to 10 sentences.

PART - A

Time : 1 hour 15 minutes

Marks : 30

3 × 2 = 6

SECTION - I

Instructions :

- (i) 3 very short answer questions are there in this **section-I**.
- (ii) Answer all the questions. Each question carries 2 marks.
- (iii) Write answer to each question in 3 to 4 sentences.

1. Write any two daily life uses of Baking soda.
2. Write the required materials, chemicals for "esterification reaction" activity

Physical Science November 20p2 English Question Paper

WJ Hussar



Physical Science November 20p2 English Question Paper:

Reviewing **Physical Science November 20p2 English Question Paper**: Unlocking the Spellbinding Force of Linguistics

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Table of Contents Physical Science November 20p2 English Question Paper

1. Understanding the eBook Physical Science November 20p2 English Question Paper
 - The Rise of Digital Reading Physical Science November 20p2 English Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science November 20p2 English Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science November 20p2 English Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science November 20p2 English Question Paper
 - Personalized Recommendations
 - Physical Science November 20p2 English Question Paper User Reviews and Ratings
 - Physical Science November 20p2 English Question Paper and Bestseller Lists

5. Accessing Physical Science November 20p2 English Question Paper Free and Paid eBooks
 - Physical Science November 20p2 English Question Paper Public Domain eBooks
 - Physical Science November 20p2 English Question Paper eBook Subscription Services
 - Physical Science November 20p2 English Question Paper Budget-Friendly Options
6. Navigating Physical Science November 20p2 English Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science November 20p2 English Question Paper Compatibility with Devices
 - Physical Science November 20p2 English Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science November 20p2 English Question Paper
 - Highlighting and Note-Taking Physical Science November 20p2 English Question Paper
 - Interactive Elements Physical Science November 20p2 English Question Paper
8. Staying Engaged with Physical Science November 20p2 English Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science November 20p2 English Question Paper
9. Balancing eBooks and Physical Books Physical Science November 20p2 English Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science November 20p2 English Question Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science November 20p2 English Question Paper
 - Setting Reading Goals Physical Science November 20p2 English Question Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science November 20p2 English Question Paper
 - Fact-Checking eBook Content of Physical Science November 20p2 English Question Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web date of publication 1943 1946 language french chronology followed by spirou et fantasio 1948 spirou et l'aventure
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web apr 19 2023 chemical reactions and equations extra questions very short answer type question 1 why is the combustion of liquified petroleum gas lpg a chemical change answer combustion of liquified petroleum gas lpg is a chemical change because after its combustion a new substance is formed and cannot be turned back into

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web q5 identify the reducing agent in the following reactions a $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$ b $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ c $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$ d $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ answer a here ammonia NH_3 is the reducing agent b here water H_2O is the reducing agent

