

Chapter 6 Chemical Bonds**Section 6.2 Covalent Bonding****(pages 165–169)**

This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.

Reading Strategy (page 165)

Relating Text and Visuals As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

Covalent Bonds (pages 165–167)

- Describe a covalent bond. _____

- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
 - electron dot
 - structural formula
 - space-filling
 - electron cloud
- Describe a molecule. _____

- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. _____
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
 - zero
 - one
 - two
 - three

Physical Science Chapter 6 Chemical Bonds Chapter Review

Hussin A.Rothana



Physical Science Chapter 6 Chemical Bonds Chapter Review:

Whispering the Secrets of Language: An Psychological Journey through **Physical Science Chapter 6 Chemical Bonds Chapter Review**

In a digitally-driven world wherever screens reign supreme and instant conversation drowns out the subtleties of language, the profound techniques and mental subtleties concealed within phrases often get unheard. However, situated within the pages of **Physical Science Chapter 6 Chemical Bonds Chapter Review** a captivating literary value blinking with natural emotions, lies a fantastic quest waiting to be undertaken. Written by a skilled wordsmith, that enchanting opus attracts viewers on an introspective journey, lightly unraveling the veiled truths and profound influence resonating within the material of each word. Within the mental depths of this touching review, we shall embark upon a honest exploration of the book is core themes, dissect its captivating publishing style, and yield to the strong resonance it evokes serious within the recesses of readers hearts.

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