

**Organic Compounds:** carbon containing chains

Alkane- single bonds

Alkene- double bonds

Alkyne- triple bonds

Naming uses prefixes based on # carbons, then the root above

Depending on bond type

1 carbon = meth-

2 carbon = eth-

3 carbon = prop-

4 carbon = but-

5 carbon = pent-

6 carbon = hex-

7 carbon = hept-

8 carbon = oct-

9 carbon = non-

10 carbon = dec-

Monomers build up to be polymers

**Nuclear Radiation:** 1) elements above atomic # 83

2) atoms with uneven #p and #n ratio

Nuclear splitting

Nuclear joining

Decay particles:

Alpha-  $\alpha$ / $^4_2\text{He}$ , positive, most massive, like helium ion

Beta-  $\beta$ / $^-$  or  $\beta$ / $^+$ , negative, neglig. Mass, like electron

Gamma-  $\gamma$ / $^0_0\gamma$ , no charge, no mass, most energy, most penetrating

Half life time for half to decay

100 gram sample has half life of 10 days. How much remains after 30

days?

30/10 = 3 half lives:  $100\text{g}/2 = 50\text{g}/2 = 25\text{g}/2 = 12.5\text{g}$ , grams remain

**Chemical Reactions:** new substance created.

**Chemical Equations**

Reactant  $\rightarrow$  Product

Synthesis:  $A + B \rightarrow AB$

Decomposition:  $AB \rightarrow A + B$

Single Displacement:  $A + BC \rightarrow AC + B$

Double Displacement:  $AB + CD \rightarrow AD + CB$

**Balanced Chemical equations:** Law of Conservation of Mass

Use coefficients- large whole numbers put front of substance

$x\text{H}_2 + \text{O}_2 \rightarrow x\text{H}_2\text{O}$

Coefficients also indicate # moles of each substance taking part in the reaction & mole ratio

Ex: 2 mol  $\text{H}_2$  react with 1 mol  $\text{O}_2$  to produce 2 mol  $\text{H}_2\text{O}$

**Molar mass:** mass of 1 mole of a substance.

Add up masses from PT for each ATOM present in substance.

Ex:  $\text{H}_2\text{O}$ : [2 atoms of H @ 1g each] + [1 atom of oxygen @ 16g each] =

18 g/mol

Created by Megan Higgins- Chemistryg

# Physical Science Eoct Review Georgia Answer Key

**AW Chickering**



## **Physical Science Eoct Review Georgia Answer Key:**

## **Physical Science Eoct Review Georgia Answer Key** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Physical Science Eoct Review Georgia Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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