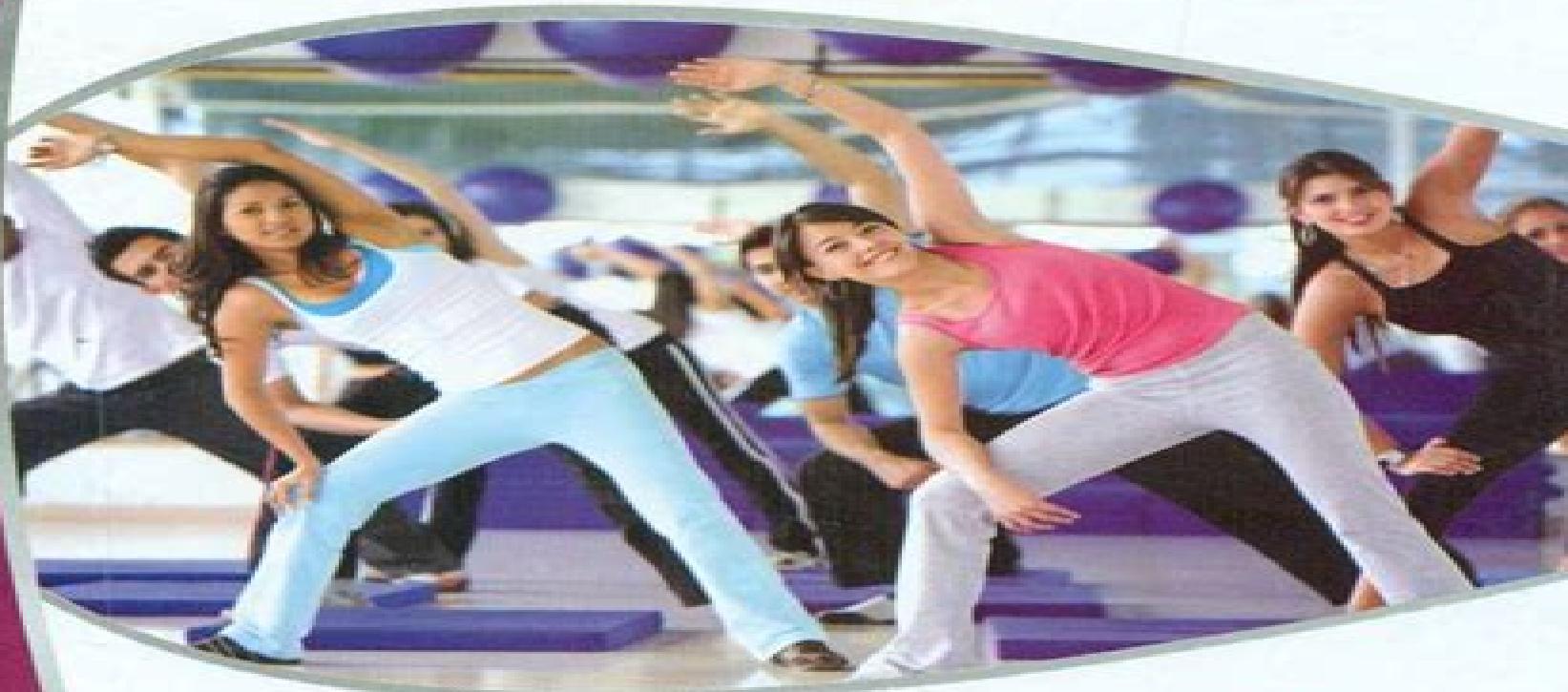




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# Physical Education Class 11 Lab Manual

**Yi-Tong Ma**



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soln 2.58 g 98.57 g 101.15 g  $\Delta T$  20.4 °C 22.5 °C 2.1 °C

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web in a calorimeter the white solid  $\text{BaSO}_4$  forms and the temperature of the mixture increases to 28.1 °C assuming that the calorimeter absorbs only a negligible quantity of heat and that the specific heat capacity of the solution is 4.18 J/g °C and that the density of the final solution is 1.0 g/mL calculate the enthalpy change per mole

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