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# OHIO PERMIT TEST 2023 REVIEW SOLUTION GUIDE

What items are checked during inspection of your car? - Steering, tires, headlights, taillights, license illuminating light, red reflectors, color of lights, turn signals, horn, safety glass, windshield wipers, rear view mirror

When your car is being inspected, how much tread depth should your tires have on the road surface? - 1/16 inch

When is it an exception to not keep to the right in the lanes? - When passing another vehicle, when driving on a road divided into three or more marked lanes, when directed by a police officer, when an obstruction makes you do so

What lane do you use if you are moving slower than the rest of traffic? - The right lane

If you are on a roadway divided into three lanes, do you drive in the center lane? - No, not unless the center lane is devoted to traffic moving in the direction the driver is proceeding

When approached by an emergency vehicle, what do you do? - Drive to a position parallel to the right edge or curb of the road or highway then stop

When an emergency vehicle is stopped alongside the road with it's lights flashing, what do you do? - The drivers must pass with care and reduce their speeds, and whenever it is possible, the driver should change lanes to pass the emergency vehicle in a lane that is not adjacent to where the emergency vehicle is.

What is important to do whenever making a right turn? - To turn right as close as possible to the right-hand curb or edge of the road

When is it legal to make a right-hand turn on a red light? - When there is no sign posted at the intersection forbidding right turns on red, you have come to a complete stop after allowing all crossing traffic and pedestrians to pass, and when you can clearly see that the turn can be made safely

What are the correct methods of making left turns? - 1. From a 4 lane with a turning lane to a 2 lane street

2. From a 2 lane to a 4 lane

3. From a 4 lane to a 2 lane

4. From a 4 lane to a 1 way street

5. From a 1 way to another 1 way

6. From a 1 way to a 4 lane

# Permit Study Guide 23

**Léo Pomar**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located on the right side of the slide, partially overlapping the light blue bar.

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web name the following molecular compounds  $\text{sc}_2$   $\text{n}_2\text{o}_4$   $\text{p}_4\text{o}_{10}$   $\text{pf}_5$   $\text{sc}_2$  sulfur dichloride  $\text{n}_2\text{o}_4$  dinitrogen tetroxide the a of tetra is dropped before the o of oxide  $\text{p}_4\text{o}_{10}$  tetraphosphorous decoxide the a of deca is dropped before the o of oxide  $\text{pf}_5$  phosphorous pentafluoride

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web examine the pairs of ternary acids in model 2 that contain sulfur phosphorus and nitrogen each pair has one acid that ends in *ic* and another that ends in *ous* these endings are related to the name of the polyatomic anion found in the acid ate or ite

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web the acids lose a hydrogen ion for each acid base reaction in model 2 describe the role of the brønsted lowry base in the proton  $H^+$  ion transfer that occurs the base gains or accepts a hydrogen ion proton as you saw in last week's pogil all arrhenius bases in model 1 have an  $OH^-$  ion in their chemical formulas

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web dec 10 2022  $HCl$  answer hydrochloric acid  $HBr$  answer hydrobromic acid  $H_2S$  answer hydrosulfuric acid  $HF$  answer hydrofluoric acid  $HClO_3$  answer chloric acid chlorate  $H_2SO_3$  answer sulfurous acid sulfite  $H_2SO_4$  answer sulfuric acid sulfate  $H_3PO_3$  answer phospho naming acids practice pogil answer key december 10

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web one of the ions produced is always h<sup>+</sup> which immediately combines with a water molecule to form the hydronium ion h<sub>3</sub>o<sup>+</sup> the h<sub>3</sub>o<sup>+</sup> ion is what defines the acidic properties of a substance because of this special classification acids have a naming system different from ionic or other molecular covalent compounds

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web the molecular or empirical formula acids have their own set of naming rules cf sec 2 8 which you must also know you also need to be able to recognize and name a few organic compounds such as simple alkanes and alcohols see sec 2 9 key questions 8 in the chemical formula and name which element is given first a metal or nonmetal 9

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