

Directions: Choose the letter that corresponds to the correct answer.

1. Which of the following can be considered as Dobereiner's triad?
 - a. He: 4.0, Li: 6.9, Na: 23.0
 - b. N: 14.0, P: 40.0, At: 217.0
 - c. Ne: 20.2, Ar: 39.9, Kr: 83.8
 - d. Ca: 40.1, Sr: 87.6, Ba: 137.3
2. Which of the following elements does not belong to the group?
 - a. N
 - b. S
 - c. Po
 - d. Te
3. Which of the following is/are TRUE about boron and oxygen?
 - a. both are metalloids
 - b. boron is smaller than oxygen
 - c. boron is more electronegative than oxygen
 - d. higher energy is required to eject an electron from the valence shell of boron
 - e. all of the above
 - f. none of the above

For 4 and 5, consider Na, Mg, Al, and Si.

4. The element that ejects one electron from its valence shell most readily is _____.
 - a. Na
 - b. Mg
 - c. Al
 - d. Si
5. Which element has the most negative electron affinity?
 - e. Na
 - f. Mg

Periodic Trends Practice Questions Answers

Rob Ritchie



Periodic Trends Practice Questions Answers:

This book delves into Periodic Trends Practice Questions Answers. Periodic Trends Practice Questions Answers is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodic Trends Practice Questions Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Periodic Trends Practice Questions Answers
- Chapter 2: Essential Elements of Periodic Trends Practice Questions Answers
- Chapter 3: Periodic Trends Practice Questions Answers in Everyday Life
- Chapter 4: Periodic Trends Practice Questions Answers in Specific Contexts
- Chapter 5: Conclusion

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