

Trends on the Periodic Table

Part 1: Identifying chemical properties/characteristics of the elements

For each of the following elements, draw a caricature or cartoon above the description to show its "chemical personality" based on the description provided. (Note: See the Neon example for ideas.)

Lithium
Being one of the smaller elements with only 3 electrons, lithium is happy to give one of its electrons away and never seems to ask for any in return.



Francium
Francium is the largest of all of the elements and has so many electrons that it has trouble keeping track of them. Francium will eagerly give up electrons to other elements and never takes in extras. But watch out, it has an extremely reactive and explosive personality!

Fluorine
One of the smallest elements, Fluorine is often described as snappy with a "magnetic" personality. As part of the halogen family (stoutest electron thieves), Fluorine is the best at convincing other elements to give up their electrons to him. However, he holds on tight to his own electrons and will not let them go without a fight.



Iodine
As part of the halogen family, Iodine is also a skilled electron thief, however, not nearly as good as its smaller brothers and sisters. Iodine will also put up a pretty good fight to keep its own electrons.



Neon
Part of the Noble family, Neon does not readily interact with other elements. With a full set of valence electrons, Neon feels that he has everything he needs to make him happy without sharing with others.

Part 2: Use your textbook to complete the following:

Atomic Radii Size:

1. Explain **atomic radius**: A measure of the size (volume) of an atom. Determined by measuring half of the distance between two identical atoms bonded together.
2. Using the information in figure 5-13 (pg 141) in your book, **write** the atomic radii size for **alkali metals & period #2** on the periodic table to your right.



3. What happens to atomic radii as you move down a column and across a row? On the arrows on the diagram, indicate the trend. (Options: gets bigger OR gets smaller)
4. Of all the elements on the periodic table, **identify & circle the largest & smallest** in size.

5. Rank the following elements from largest to smallest in atomic radii:

a) Be, Ba, Mg = **Ba, Mg, Be**

b) Li, F, B = **Li, B, F**

6. How can you explain the trend?

a. Why do they get smaller as you go across a period? Additional protons in the nucleus "pull" on electrons, bring the electron cloud closer.

Effective nuclear charge, Z_{eff} , increases the attraction of the nucleus and therefore pulls the electron cloud closer to the nucleus resulting in smaller radii.

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JR Anderson



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The Top Books of the Year Periodic Trends Pogil Worksheet Answers Extended Answer The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have charmed audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and suspenseful novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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Table of Contents Periodic Trends Pogil Worksheet Answers Extended Answer

1. Understanding the eBook Periodic Trends Pogil Worksheet Answers Extended Answer
 - The Rise of Digital Reading Periodic Trends Pogil Worksheet Answers Extended Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Pogil Worksheet Answers Extended Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Pogil Worksheet Answers Extended Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Pogil Worksheet Answers Extended Answer
 - Personalized Recommendations
 - Periodic Trends Pogil Worksheet Answers Extended Answer User Reviews and Ratings
 - Periodic Trends Pogil Worksheet Answers Extended Answer and Bestseller Lists
5. Accessing Periodic Trends Pogil Worksheet Answers Extended Answer Free and Paid eBooks
 - Periodic Trends Pogil Worksheet Answers Extended Answer Public Domain eBooks
 - Periodic Trends Pogil Worksheet Answers Extended Answer eBook Subscription Services
 - Periodic Trends Pogil Worksheet Answers Extended Answer Budget-Friendly Options
6. Navigating Periodic Trends Pogil Worksheet Answers Extended Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Pogil Worksheet Answers Extended Answer Compatibility with Devices
 - Periodic Trends Pogil Worksheet Answers Extended Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Pogil Worksheet Answers Extended Answer
 - Highlighting and Note-Taking Periodic Trends Pogil Worksheet Answers Extended Answer
 - Interactive Elements Periodic Trends Pogil Worksheet Answers Extended Answer
8. Staying Engaged with Periodic Trends Pogil Worksheet Answers Extended Answer

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Pogil Worksheet Answers Extended Answer
9. Balancing eBooks and Physical Books Periodic Trends Pogil Worksheet Answers Extended Answer
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Pogil Worksheet Answers Extended Answer
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Pogil Worksheet Answers Extended Answer
- Setting Reading Goals Periodic Trends Pogil Worksheet Answers Extended Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Pogil Worksheet Answers Extended Answer
- Fact-Checking eBook Content of Periodic Trends Pogil Worksheet Answers Extended Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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