Use the information in these tables to complete the graph as described below

| Symbol | Atomic Radius (Picometers) | First Ionization energy (kilojoules/mole) | Electronegativity (4-point scale) 2.1 1.0 0.9 | |
|--------|----------------------------------|---|---|--|
| H | 31 | 1312 | | |
| IL-i | 128 | 520 | | |
| Nia | 166 | 496 | | |
| K | 203 | 410 | 0.8 | |
| Rb | 220 | 403 | 0.8 | |
| Cs | 244 | 376 | 0.7 | |

| Symbol | Atomic Radius (Picometers) | First lonization energy (kilojoules/mole) | Electronegativity (4-point scale) 0.9 1.2 | | |
|--------|----------------------------------|---|--|--|--|
| Na | 166 | 496 | | | |
| Mg | 141 | 738 | | | |
| AL | 121 | 578 | 1.5 | | |
| Si | 111 | 787 | | | |
| P | 107 | 1012 | 2.1 | | |
| S | 105 | 1000 | 2.5 | | |
| CL | 102 | 1251 | 3.0 | | |
| Ar | 106 | 1521 | | | |

- Using colored pencils or pen, list under the symbol (in this order!) the Atomic Radius, First Ionization Energy, and Electronegativity.
- Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First
 ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are
 available, as long as you are consistent.
- 3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
- Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)

| Li | Across a period (→) first ionization energy tends to | | | | | | | |
|-------------------------|--|------------|--------------------------------|--------------|---|----|---|--|
| Na 166 496 0.9 | Mg | Al | Si | P | s | CI | A | |
| К | Down | a Group (』 |) atomic rac | lius tends t | o | | | |
| Rb | 01,000,000,000 | |) first ioniza) electroneç | | | | | |

Periodic Trends Activity Packet Answers

Kim Cooper, Kelly Gosnell

Periodic Trends Activity Packet Answers:

Principles of Epidemiology Workbook: Exercises and Activities Ray M. Merrill, 2010-09-15 This workbook was written for students of epidemiology and serves as a supplement to any one of several introductory text books in epidemiology Each chapter is divided into an introduction a series of guestions and detailed responses and a series of Homework guestions At the end of each chapter is a table with a list of selected epidemiology text books with accompanying chapters in those books that the workbook chapter may supplement The general learning outcomes LOs for this workbook are 1 Become familiar with basic concepts and definitions commonly used in epidemiology 2 Define a public health problem 3 Identify appropriate uses and limitations of data and research design strategies for solving public health problems 4 Make relevant inferences from quantitative and qualitative data 5 Distinguish between statistical association and cause effect relationships 6 Measure and describe patterns of disease incidence prevalence and mortality 7 Identify environmental factors and behaviors associated with health related states or events 8 Be familiar with the steps for investigating disease outbreaks 9 Identify calculate and interpret common indices used in identifying the health status 10 Evaluate program effectiveness 11 Critically assess epidemiological research 12 Be able to communicate health findings Each chapter features 10 20 mastery check questions with detailed answers 5 optional problems A case study A multiple choice short answer guiz Answers to the cases and guizzes are provided as part of the online instructor resource package **Hands-On General Science Activities With Real-Life** Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life Class 8-12 Chemistry Questions and Answers PDF Arshad Igbal, The Class 8 12 Chemistry Quiz Questions and Answers PDF Grade 8 12 Chemistry Competitive Exam Questions Chapter 1 15 Practice Tests Chemistry Textbook Questions for Beginners includes Questions to solve problems with hundreds of class questions Class 8 12 Chemistry Questions and Answers PDF book covers basic concepts and analytical assessment tests Class 8 12 Chemistry Quiz PDF book helps to practice test questions from exam prep notes The Grade 8 12 Chemistry Quiz Questions and Answers PDF eBook includes Practice material with verbal quantitative and analytical past papers questions Class 8 12 Chemistry Questions and Answers PDF Free download chapter 1 a book to review textbook questions on chapters Molecular structure acids and bases atomic structure bonding chemical equations descriptive chemistry equilibrium systems gases laboratory liquids and solids mole concept oxidation reduction rates of reactions solutions thermochemistry Questions for high school and college revision questions Chemistry Interview Questions and Answers PDF Download free eBook's sample covers beginner's solved questions textbook's study notes to practice online tests The Grade 8 12 Chemistry Interview Questions Chapter 1 15 PDF book includes high school workbook questions to

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