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what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is a helpful way to remember this

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 oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3 \text{ Ag}$ oxidation reduction $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 O_2

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 $\text{a } \text{I}_2 + \text{O}_5 \rightarrow \text{I}^{2+} + \text{O}^{5-}$
 $5\text{Co} + \text{I}_2 \rightarrow 5\text{Co}^{2+} + \text{I}_2$ $5 \text{ Co} + \text{I}_2 \rightarrow 5 \text{ Co}^{2+} + \text{I}_2$

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