Student Name:	Print.	Date
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#### Periodic Table Puzzle

						1	

Using the following hat of class, place each element in its proper position on the periodic table. The letters bear no similarity to the octual symbols for the elements. The trumpated periodic table above these only the first four periods of the representative elements and the nable gases. Hint - place each group in its proper position first, and then arrange the elements in each group.

#### CLUES

- The following elements are grouped together into faculties: ERH, GSI, NLW, FQK, CDAB, UFZ, YXJ, and MYOT.
- C is a more reactive metal than A and B.
- O forms an oxide with oxygen in the following manner: Ox0s.
- 4. If is used as a medication for manic depression (bipolar disorder).
- Thus the highest mass on the periodic table.
- 6. Zisa liquid.
- I is a non-mutal that was originally entracted from uring.
- 8. If has three valence electrons on the third energy level and can be found in the kitchen.
- 9. G has a higher electron affinity than both Y and K.
- 10. S is a semi-metal that is used in the manufacturing of computer chips.
- 11. It is the largest atom of the periodic table.
- 12. N has a smaller I" ionization energy than P or X.
- 13. V does not have an electronogativity value.
- 14. Y is more dense than I and X
- 15. Q has a senaller ionization energy than E.
- 16. Whas a greater effective nuclear charge than L but less than N.
- 17. Hi reacts with water to form a white, milky base solution and hydrogen gas.
- 18. A is an element which can form both an axion and cation.
- S loses all of its 3" shell electrons to form a +4 charge.
- 20. P is the most electrosegative element on the periodic table.
- 21. O is lighter than air and has the lowest moiting point of all the elements.
- 22. U has a more negotive electron affinity than both M and W.
- 23. R is larger than E but smaller than H.
- 24. G is named after a country in Europe.
- 25. N is the only member of its group that is a gas.

# **Periodic Puzzle Answer Key**

**C** Cleary

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