

Fictitious symbols are used for the first 18 elements in the periodic table. Use the clues below to write the fictitious symbol in the appropriate spot on the periodic table provided. Symbols for real elements do not represent those elements. HINT: You do not have to complete each clue in order.

| 1 | 2 | 13            | 14            | 15             | 16 | 17 | 18 |
|---|---|---------------|---------------|----------------|----|----|----|
| X |   | A             | T             | H <del>P</del> | G  | Q  | L  |
| U | R | <del>AI</del> | <del>DP</del> | Z              | E  | K  | W  |
| J | M |               |               |                |    |    | V  |

- Clue #1      U and J are alkali metals. J has more energy levels than U.
- Clue #2      I has 4 valence electrons on the ~~3<sup>rd</sup>~~<sup>2<sup>nd</sup></sup> energy level.
- Clue #3      M is a metal in period 3 with 2 valence electrons
- Clue #4      X has one proton in its nucleus
- Clue #5      Q has 2 energy levels and is a halogen
- Clue #6      L is the only noble gas that doesn't have 8 valence electrons
- Clue #7      A is the only metalloid in the 13<sup>th</sup> group.
- Clue #8      I has 3 valence electrons on the third energy level
- Clue #9      V and W are un-reactive elements. V has more electrons than W.
- Clue #10     E and G have a common ion of -2 ( $E^{-2}$ ,  $G^{-2}$ ). E has 16 neutrons.
- Clue #11     R has an nuclear symbol of  ${}^3_1R$
- Clue #12     K has one less electron than V
- Clue #13     D and Z are in the same period as J. However D is a metalloid and Z is a nonmetal.
- Clue #14     H has 7 neutrons

# Periodic Puzzling Answer Key

**Clemens Wendtner**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Periodic Puzzling Answer Key:**

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