

# Periodic Trends Worksheet

Name: \_\_\_\_\_ Grade: \_\_\_\_\_ ID: \_\_\_\_\_

**Directions:** Use your notes to answer the following questions.

- 1 Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

**Ans:** O, C, Al, K

- 2 Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

**Ans:** Ne, Al, S, O

- 3 Why does fluorine have a higher ionization energy than iodine?

**Ans:** Fluorine has a smaller atomic radius, so the proton can exert a greater pull on electron (Valence shell is close)

- 4 Why do elements in the same family generally have similar properties?

**Ans:** They have the same valence shell electron arrangement.

- 5 Indicate whether the following properties increase or decrease from left to right across the periodic table.

- a. atomic radius (excluding noble gases) **Decrease**
- b. first ionization energy **Increase**
- c. electronegativity **Increase**

- 6 What trend in atomic radius occurs across the periodic table? What causes this trend?

**Ans:** Atomic radius increases down a group because energy levels (shells) are added.

- 7 What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

**Ans:** Ionization energy increases across a period because as elements become less metallic, it requires more energy to remove an electron.

- 8 Circle the atom in each pair that has the largest radius.

a. Al or B

d. O or F

b. Na or Al

e. Br or Cl

c. S or O

f. Mg or Ca

# Periodic Table Trends Answer Key

**The Princeton Review**



## **Periodic Table Trends Answer Key:**

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Constructing Measures Mark Wilson,2004-12-13 Constructing Measures introduces a way to understand the advantages and disadvantages of measurement instruments how to use such instruments and how to apply these methods to develop new instruments or adapt old ones The book is organized around the steps taken while constructing an instrument It opens with a summary of the constructive steps involved Each step is then expanded on in the next four chapters These chapters develop the building blocks that make up an instrument the construct map the design plan for the items the outcome space and the statistical measurement model The next three chapters focus on quality control They rely heavily on the calibrated construct map and review how to check if scores are operating consistently and how to evaluate the reliability and validity evidence The book introduces a variety of item formats including multiple choice open ended and performance items projects portfolios Likert and Guttman items behavioral observations and interview protocols Each chapter includes an overview of the key concepts related resources for further investigation and exercises and activities Some chapters feature appendices

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**Activities With Real-Life Applications** Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life [Chemistry in the Community \(ChemCom\)](#) American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom s wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards **Chemistry in a Month** John Wilkinson, 2003 **Shriver and Atkins' Inorganic Chemistry** Peter Atkins, 2010 Inorganic Chemistry fifth edition represents an integral part of a student s chemistry education Basic chemical principles are set out clearly in Foundations and are fully developed throughout the text culminating in the cutting edge research topics of the Frontiers which illustrate the dynamic nature of inorganic chemistry

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**Inorganic Chemistry** Mark Weller, Mark T. Weller, Tina Overton, Jonathan Rourke, Fraser Armstrong, 2014 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject Inorganic Chemistry Sixth Edition is the ideal course companion for the duration of a student's degree The authors have drawn upon their extensive teaching and research experience in updating this established text the sixth edition retains the much praised clarity of style and layout from previous editions while offering an enhanced Frontiers section Exciting new applications of inorganic chemistry have been added to this section in particular relating to materials chemistry and medicine This edition also sees a greater use of learning features to provide students with all the support they need for their studies Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resource Centre For registered adopters of the text Figures marginal structures and tables of data ready to download Test bank For students Answers to self tests and exercises from the book Videos of chemical reactions Tables for group theory Web links Interactive structures and other resources on [www.chemtube3d.com](http://www.chemtube3d.com)

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**AP Chemistry Premium, 2025: Prep Book with 6 Practice Tests + Comprehensive Review + Online Practice** Neil D. Jespersen, Pamela Kerrigan, 2024-07-02. Be prepared for exam day with Barron's Trusted content from AP experts. Barron's AP Chemistry Premium 2025 includes in-depth content review and practice. It's the only book you'll need to be prepared for exam day. Written by Experienced Educators. Learn from Barron's: all content is written and reviewed by AP experts. Build your understanding with comprehensive review tailored to the most recent exam. Get a leg up with tips, strategies, and study advice for exam day; it's like having a trusted tutor by your side. Be Confident on Exam Day. Sharpen your test-taking skills with 6 full-length practice tests: 3 in the book and 3 more online, plus 3 short diagnostic tests for assessing strengths and areas for improvement and detailed answer explanations for all questions. Strengthen your knowledge with in-depth review covering all units on the AP Chemistry exam. Reinforce your learning with more than 300 practice questions throughout the book that cover all frequently tested topics. Learn what to expect on test day with essential details about the exam format, scoring, calculator policy, strategies for all question types, and advice for developing a study plan. Robust Online Practice. Continue your practice with 3 full-length practice tests on Barron's Online Learning Hub. Simulate the exam experience with a timed test option. Deepen your understanding with detailed answer explanations and expert advice. Gain confidence with scoring to check your learning progress. Power up your study sessions with Barron's AP Chemistry on Kahoot! additional free practice to help you ace your exam.

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