

# Periodic Table Notes:

## Determining # of Shells and Valence Electrons

### Directions:

View the Power Point study guide with the same title as this handout. Record all your answers below.

1. In the periodic table, each row is called a \_\_\_\_\_.
2. The elements in each \_\_\_\_\_ have the same number of \_\_\_\_\_.
3. Each column in the periodic table is called a \_\_\_\_\_.
4. Each of the elements in the same \_\_\_\_\_ have the same number of \_\_\_\_\_ in their outer shells.
5. The electrons in the outer shell are called \_\_\_\_\_ electrons.
6. The \_\_\_\_\_ number equals the \_\_\_\_\_ of \_\_\_\_\_ electrons.

Complete the following data table as you view the rest of the slides.

Symbol	Element	Period	# of Shells	Group #	# Valence electrons
C					
Na					
Ne					
H					
Be					
S					
K					
He					

# Periodic Table Study Guide Worksheet

**Ian Pickup**



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