

Periodic Table of the Elements

The periodic table is color-coded by groups and periods. The groups are labeled at the top with their respective valence electron counts: +1, +2, +3, n/e, -3, -2, -1, and 0. The periods are labeled on the left from 1 to 7. The elements are arranged in a grid, with the lanthanide and actinide series shown separately at the bottom.

The only metal that is liquid at room temperature.

Period 3 element with 7 valence electrons.

Semi-conductor metalloids used in electronics.

Only one valence electron, but not part of a family.

Extremely reactive non-metal.

Group that will form 2+ cations.

The only non-metal that is liquid at room temperature.

Completely un-reactive family with complete valence shell.

Extremely reactive metal.

Group with 4 valence electrons.

Periodic Trends Review Answers

L Cohen



Periodic Trends Review Answers:

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