

Periodic Table of the Elements

Periodic Table of the Elements

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1	H 1.008	He 4.003											Li 6.941	Be 9.012								B 10.811	C 12.011	N 14.007	O 15.999	F 18.998	Ne 20.180	
2	Na 22.990	Mg 24.305											Al 26.982	Si 28.086	P 30.974	S 32.06	Cl 35.45	Ar 39.948										
3	K 39.098	Ca 40.078	Sc 44.956	Ti 47.88	V 50.942	Cr 52.00	Mn 54.938	Fe 55.845	Co 58.933	Ni 58.69	Cu 63.546	Zn 65.38	Ga 69.723	Ge 72.64	As 74.922	Se 78.96	Br 79.904	Kr 83.80										
4	Rb 85.468	Sr 87.62	Y 88.906	Zr 91.224	Nb 92.906	Mo 95.94	Tc 98.906	Ru 101.07	Rh 102.91	Pd 106.42	Ag 107.87	Cd 112.41	In 114.82	Sn 118.71	Sb 121.76	Te 127.6	I 126.905	Xe 131.29										
5	Cs 132.905	Ba 137.327	La 138.905	Ce 140.12	Pr 140.908	Nd 144.24	Pm 144.913	Sm 150.36	Eu 151.964	Gd 157.25	Tb 158.925	Dy 162.50	Ho 164.930	Er 167.259	Tm 168.930	Yb 173.054	Lu 174.967											
6	Po 209	At 210	Rn 222	Fr 223	Ra 226	Ac 227	Th 232.0377	Pa 231.03688	U 238.02891	Np 237.04817	Pu 244.0642	Am 243.06138	Cm 247.07035	Bk 247.07035	Cf 251.0825	Es 252.083	Fm 257	Mendelevium 258	Nobelium 259	Lr 262								
7	Uu 288	Uub 293	Uut 289	Uuq 294	Uup 289	Uuh 295	Uus 294	Uuo 294	Uut 294	Uuh 295	Uus 296	Uuo 297	Uut 297	Uuh 298	Uus 299	Uuo 301	Uut 301	Uuh 302	Uus 303	Uuo 304								

The only metal that is liquid at room temperature.

Period 3
element with
7 valence
electrons.

Semi-conductor
metalloid used
in electronics.

Only one valence electron, but not part of a family.

Extremely reactive non-metal.

Group that will form 2+ cations.

The only non-metal that is liquid at room temperature.

Completely
un-reactive family
with complete
valence shell.

Extremely reactive metal.

Group with 4 valence electrons.

Periodic Trends Review

DJ Losen



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