

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
U.S. Department of Commerce

		Frequently used fundamental physical constants										Physics Laboratory		Standard Reference Data						
		For the most accurate values of these and other constants, visit physics.nist.gov/constants										physics.nist.gov		www.nist.gov/srd						
		1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																		
		speed of light in vacuum c = 299 792 458 m s ⁻¹ (exact)																		
		Planck constant h = 6.626 070 15 × 10 ⁻³⁴ J s (exact)																		
		elementary charge e = 1.602 176 634 × 10 ⁻¹⁹ C																		
		electron mass m _e = 9.109 383 56 × 10 ⁻³¹ kg																		
		proton mass m _p = 1.672 621 9 × 10 ⁻²⁷ kg																		
		fine-structure constant α = 1/137.036																		
		Rydberg constant R _∞ = 10 973 731.568 162 m ⁻¹																		
		Bohrmann constant k _B = 1.380 658 × 10 ⁻²³ J K ⁻¹																		

Periodic Table Chemistry If8766

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Monaghan**



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