

Name: _____

Dates:

CHEMISTRY

Periodic Table of the Elements

Use the Periodic Table of the Elements to answer the questions.

The periodic table is color-coded by groups:

- Alkali metal (Orange)
- Alkaline earth metal (Yellow)
- Metal (Light Green)
- Transition metal (Dark Green)
- Lanthanide (Teal)
- Metalloid (Light Blue)
- Non-metal (Light Purple)
- Halogen (Dark Purple)
- Noble gas (Dark Blue)
- Radioactive (Pink)

Key elements and groups are labeled:

- Hydrogen (H)
- Lithium (Li)
- Beryllium (Be)
- Sodium (Na)
- Magnesium (Mg)
- Potassium (K)
- Calcium (Ca)
- Scandium (Sc)
- Titanium (Ti)
- Vanadium (V)
- Chromium (Cr)
- Manganese (Mn)
- Iron (Fe)
- Cobalt (Co)
- Nickel (Ni)
- Copper (Cu)
- Zinc (Zn)
- Gallium (Ga)
- Germanium (Ge)
- Arsenic (As)
- Selenium (Se)
- Bromine (Br)
- Krypton (Kr)
- Rubidium (Rb)
- Sr
- Y
- Zr
- Nb
- Mo
- Tc
- Ru
- Rh
- Pd
- Ag
- Cd
- In
- Sn
- Sb
- Te
- I
- Xe
- Cs
- Ba
- La-Lu
- Hf
- Ta
- W
- Re
- Os
- Ir
- Pt
- Au
- Hg
- Tl
- Pb
- Bi
- Po
- At
- Rn
- Fr
- Ra
- Ac-Lr
- Rf
- Db
- Sg
- Bh
- Hs
- Mt
- Ds
- Rg
- Cn
- Nh
- Fl
- Mc
- Lv
- Ts
- Og

Other labels include:

- Atomic number (5, 10, 811)
- Symbol (B)
- Name (Boron)
- Group (13)
- Period (2)
- Block (p)
- Category (Metalloid)

Part 1: Atomic Number. On the line next to the element, write the atomic number of the element. Your answer should be 1, 2, or 3 digits. No letters.

- | | | | | | |
|----|--|-----------|-----|--|--------------|
| 1. | | Hydrogen | 7. | | Cerium |
| 2. | | Oxygen | 8. | | Iridium |
| 3. | | Magnesium | 9. | | Bismuth |
| 4. | | Vanadium | 10. | | Uranium |
| 5. | | Krypton | 11. | | Fermium |
| 6. | | Ruthenium | 12. | | Darmstadtium |

Periodic Table Answers Section 3

Michael Hoelscher

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Periodic Table Answers Section 3

1. Understanding the eBook Periodic Table Answers Section 3
 - The Rise of Digital Reading Periodic Table Answers Section 3
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Answers Section 3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Answers Section 3
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Answers Section 3

- Personalized Recommendations
- Periodic Table Answers Section 3 User Reviews and Ratings
- Periodic Table Answers Section 3 and Bestseller Lists
- 5. Accessing Periodic Table Answers Section 3 Free and Paid eBooks
 - Periodic Table Answers Section 3 Public Domain eBooks
 - Periodic Table Answers Section 3 eBook Subscription Services
 - Periodic Table Answers Section 3 Budget-Friendly Options
- 6. Navigating Periodic Table Answers Section 3 eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Answers Section 3 Compatibility with Devices
 - Periodic Table Answers Section 3 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Answers Section 3
 - Highlighting and Note-Taking Periodic Table Answers Section 3
 - Interactive Elements Periodic Table Answers Section 3
- 8. Staying Engaged with Periodic Table Answers Section 3
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Answers Section 3
- 9. Balancing eBooks and Physical Books Periodic Table Answers Section 3
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Answers Section 3
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Answers Section 3
 - Setting Reading Goals Periodic Table Answers Section 3
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Answers Section 3

- Fact-Checking eBook Content of Periodic Table Answers Section 3
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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