

Saturation refers to any Gas - Vapor Combination. While -
 • Humidity refers only Air - Water Vapor System.

Absolute Humidity It is defined as the weight of Water Vapour present in a unit weight of dry (Non - condensable) gas.

$$H = \frac{\text{kg of water Vapour}}{\text{kg of dry Air}}$$

$$\begin{array}{l} \text{Vapor} = \text{Condensed} \\ \text{gas} = \text{Liquified} \end{array}$$

⇒ Molal Humidity (Molal Saturation) It is defined as the ratio of Mole of Vapour [Condensable] to the Mole of dry (Non - condensable) gas.

$$H_m = \frac{\text{kmol of Vapor}}{\text{kmol of dry gas}}$$

for air - water vapor system

$$H = H_m \left(\frac{18}{29} \right) = 0.6207 H_m$$

In Vapor - gas Mixture

$$P = P_A + P_B$$

$$PV = nRT \Rightarrow \left. \begin{array}{l} P_A = \frac{n_A}{n_A + n_B} P \\ P_B = \frac{n_B}{n_A + n_B} P \end{array} \right\}$$

P = total / Barometric pr.

P_A = Partial pr. of Vapor

P_B = Partial pr. of ~~gas~~ gas

$$\Rightarrow \left[\frac{P_A}{P_B} = \frac{n_A}{n_B} \right]$$

then

$$H_m = \frac{P_B}{P_B} = \frac{P_A}{P - P_A} \Rightarrow$$

$$H = 0.6207 \frac{P_A}{P - P_A}$$

Process Calculation Chemical Engineering

Robert R. Redfield



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Table of Contents Process Calculation Chemical Engineering

1. Understanding the eBook Process Calculation Chemical Engineering
 - The Rise of Digital Reading Process Calculation Chemical Engineering
 - Advantages of eBooks Over Traditional Books
2. Identifying Process Calculation Chemical Engineering
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Process Calculation Chemical Engineering
 - User-Friendly Interface
4. Exploring eBook Recommendations from Process Calculation Chemical Engineering
 - Personalized Recommendations

- Process Calculation Chemical Engineering User Reviews and Ratings
- Process Calculation Chemical Engineering and Bestseller Lists
- 5. Accessing Process Calculation Chemical Engineering Free and Paid eBooks
 - Process Calculation Chemical Engineering Public Domain eBooks
 - Process Calculation Chemical Engineering eBook Subscription Services
 - Process Calculation Chemical Engineering Budget-Friendly Options
- 6. Navigating Process Calculation Chemical Engineering eBook Formats
 - ePub, PDF, MOBI, and More
 - Process Calculation Chemical Engineering Compatibility with Devices
 - Process Calculation Chemical Engineering Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Process Calculation Chemical Engineering
 - Highlighting and Note-Taking Process Calculation Chemical Engineering
 - Interactive Elements Process Calculation Chemical Engineering
- 8. Staying Engaged with Process Calculation Chemical Engineering
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Process Calculation Chemical Engineering
- 9. Balancing eBooks and Physical Books Process Calculation Chemical Engineering
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Process Calculation Chemical Engineering
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Process Calculation Chemical Engineering
 - Setting Reading Goals Process Calculation Chemical Engineering
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Process Calculation Chemical Engineering
 - Fact-Checking eBook Content of Process Calculation Chemical Engineering

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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