

REPORT SHEET

Name: _____
 Course & Year: PSYCHOLOGY - 1A
 Professor's Signature: [Signature]

Rating: 93%
 Date: November 12, 2018
 Group #: 2 - LECTURE

Experiment No. _____ STOICHIOMETRY

"Success is just a word unless you make it happen." -Dr. Naj

- I. Objective: Determine the relationship between the amount of the reactants used and the products formed in a chemical reaction.
- II. Observation and Results:

Test tube no.	No. of drops of 1 M $Pb(NO_3)_2$	No. of drops of 1 M $Pb(NO_3)_2$	Height of ppt (mm)
1.	5 drops	5 drops	0.1
2.	10 drops	10 drops	0.2
3.	15 drops	15 drops	0.3
4.	20 drops	20 drops	0.4
5.	25 drops	25 drops	0.5

QUESTIONS

1. Write the balanced equation for the reaction that took place. Indicate the gram-formula weight of the substances involved under their corresponding formulae in the equation.

$Pb(NO_3)_2 + NaCl \rightarrow PbCl_2 + 2NaNO_3$
327.2 58.5 278.1 233.0
278.1 58.5 278.1 233.0
278.1 58.5 278.1 233.0

2. How do you explain the shape of the curve obtained when the amount of $Pb(NO_3)_2$ solution was plotted against the height of the precipitate?

The shape of the curve obtained is linear, which shows that as the amount of $Pb(NO_3)_2$ increases, the height of the precipitate also increases.

The graph shows a linear relationship between the height of the precipitate and the amount of $Pb(NO_3)_2$ solution used, which is in accordance with the stoichiometry of the reaction.

"What you focus on, expands." -Jim Rohn

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