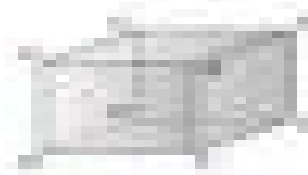


Answer following



- 1. Substrate and temperature: 20°C 30°C 40°C
- 2. A graph of product formed: product formed time min
- 3. A graph of initial rate: initial rate time min

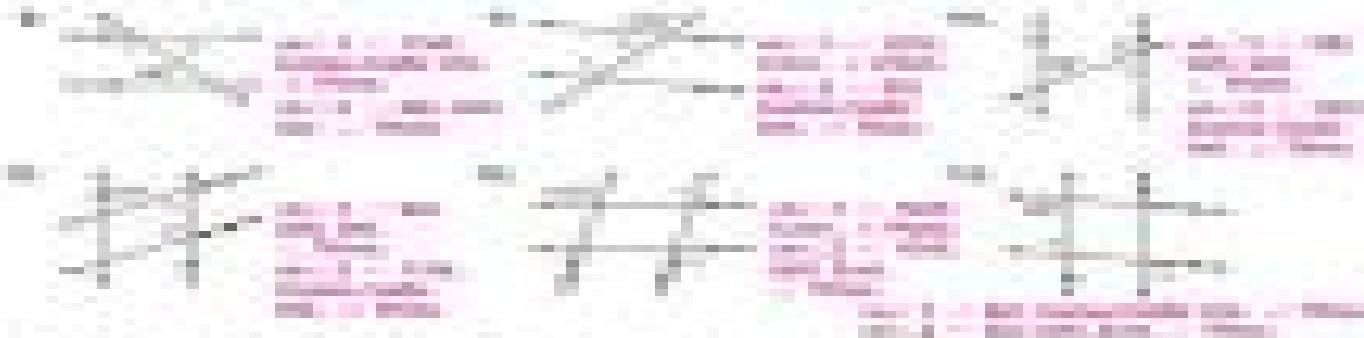
Examine all graphs of rate & time of reaction and determine the effect of temperature on the rate of reaction.

- 1. increasing temperature: 10°C 20°C 30°C 40°C
- 2. increasing temperature: 10°C 20°C 30°C 40°C
- 3. increasing temperature: 10°C 20°C 30°C 40°C
- 4. increasing temperature: 10°C 20°C 30°C 40°C



Answer 10/10

Graphs of rate of reaction vs. time. The graphs show the effect of temperature on the rate of reaction.



Graphs of rate of reaction vs. time. The graphs show the effect of temperature on the rate of reaction.



Graphs of rate of reaction vs. time. The graphs show the effect of temperature on the rate of reaction.

Prentice Hall Foundations Geometry Worksheet Answers

Robert R. Redfield



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