

Study Guide

Key Concepts



14.1 Properties of Gases

- Gases are easily compressed because of the space between particles in a gas.
- The amount of gas (n), volume (V), and temperature (T) are factors that affect gas pressure (P).



14.2 The Gas Laws

- As the pressure of a gas increases, the volume decreases, if the temperature is constant.
- As the temperature of an enclosed gas increases, the volume increases, if the pressure is constant.
- As the temperature of an enclosed gas increases, the pressure increases, if the volume is constant.

- The combined gas law allows you to do calculations for situations in which only the amount of gas is constant.



14.3 Ideal Gases

- To calculate the number of moles of a contained gas requires an expression that contains the variable n .
- Real gases differ most from an ideal gas at low temperatures and high pressures.



14.4 Gases: Mixtures and Movements

- In a mixture of gases, the total pressure is the sum of the partial pressures of the gases.
- Gases of lower molar mass diffuse and effuse faster than gases of higher molar mass.

Vocabulary

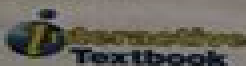
- Boyle's law (p. 418)
- Charles's law (p. 420)
- combined gas law (p. 424)
- compressibility (p. 413)
- Dalton's law of partial pressures (p. 432)
- diffusion (p. 435)
- effusion (p. 435)
- Gay-Lussac's law (p. 422)
- Graham's law of effusion (p. 435)
- ideal gas constant (p. 426)
- ideal gas law (p. 426)
- partial pressure (p. 432)

Key Equations

- Boyle's law: $P_1 \times V_1 = P_2 \times V_2$
- Charles's law: $\frac{V_1}{T_1} = \frac{V_2}{T_2}$
- Gay-Lussac's law: $\frac{P_1}{T_1} = \frac{P_2}{T_2}$
- Combined gas law: $\frac{P_1 \times V_1}{T_1} = \frac{P_2 \times V_2}{T_2}$
- Ideal gas law: $P \times V = n \times R \times T$ or $PV = nRT$
- Dalton's law: $P_{\text{total}} = P_1 + P_2 + P_3 + \dots$
- Graham's law: $\frac{\text{Rate}_A}{\text{Rate}_B} = \sqrt{\frac{\text{molar mass}_B}{\text{molar mass}_A}}$

Organizing Information

Use these terms to construct a concept map that organizes the major ideas of this chapter.



Concept Map 14 Solve the Concept Map with the help of an interactive guided tutorial.

with **ChemASAP**

Boyle's law

Charles's law

partial pressure

combined gas law

diffusion

Prentice Hall Chemistry Section 14 Review Answers

A. Berka,J. Vulterin,J. Zýka



Prentice Hall Chemistry Section 14 Review Answers:

Prentice Hall Chemistry, 2000 Modern Enolate Chemistry Manfred Braun, 2016-02-23 Authored by one of the world's leading synthetic chemists in the field this reference presents modern enolate chemistry with an emphasis on metal O enolates in asymmetric synthesis While great care is taken to cover novel successful concepts such classical methods as the famous Evans enolates are equally highlighted Throughout the book representative reaction procedures are presented thus helping readers to find the best solution for their own synthetic problem Of high interest to synthetic chemists in academia as well as the pharmaceuticals agrochemicals and fine chemicals industries **Basic Chemistry** William S. Seese, Guido H. Daub, 1985 *Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science*, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Applied Mechanics Reviews*, 1987 **Introduction to Quantum Mechanics in Chemistry** Mark A. Ratner, George C. Schatz, 2001 This book serves as a self study guide to familiarize users with the crucial language of modern chemistry science It provides a background of electronic structure programs and includes worked examples in problem solving and computer exercises For computational chemists materials scientists and chemical engineers who want to learn more about their field without unnecessary complexity detail or formalism *Teaching Chemistry - A Studybook* Ingo Eilks, Avi Hofstein, 2013-04-20 This book focuses on developing and updating prospective and practicing chemistry teachers pedagogical content knowledge The 11 chapters of the book discuss the most essential theories from general and science education and in the second part of each of the chapters apply the theory to examples from the chemistry classroom Key sentences tasks for self assessment and suggestions for further reading are also included The book is focused on many different issues a teacher of chemistry is concerned with The chapters provide contemporary discussions of the chemistry curriculum objectives and assessment motivation learning difficulties linguistic issues practical work student active pedagogies ICT informal learning continuous professional development and teaching chemistry in developing environments This book with contributions from many of the world's top experts in chemistry education is a major publication offering something that has not previously been available Within this single volume chemistry teachers teacher educators and prospective teachers will find information and advice relating to key issues in teaching such as the curriculum assessment and so forth but contextualised in terms of the specifics of teaching and learning of chemistry and drawing upon the extensive research in the field Moreover the book is written in a scholarly style with extensive citations to the literature thus providing an excellent starting point for teachers and research students undertaking scholarly studies in chemistry education whilst at the same time offering insight and practical advice to support the planning

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Warner,2012-05-14 Intended as a textbook for courses involving preparative solid state chemistry this book offers clear and detailed descriptions on how to prepare a selection of inorganic materials that exhibit important optical magnetic and electrical properties on a laboratory scale The text covers a wide range of preparative methods and can be read as separate independent chapters or as a unified coherent body of work Discussions of various chemical systems reveal how the properties of a material can often be influenced by modifications to the preparative procedure and vice versa References to mineralogy are made throughout the book since knowledge of naturally occurring inorganic substances is helpful in devising many of the syntheses and in characterizing the product materials A set of questions at the end of each chapter helps to connect theory with practice and an accompanying solutions manual is available to instructors This book is also of appeal to postgraduate students post doctoral researchers and those working in industry requiring knowledge of solid state synthesis

Newer Redox Titrants A. Berka, J. Vulterin, J. Zýka, 2013-10-22 International Series of Monographs in Analytical Chemistry Volume 22 Newer Redox Titrants focuses on the processes reactions methodologies and approaches involved in the study of redox titrants The publication first offers information on potassium permanganate in alkaline solution and compounds of trivalent manganese including standard solutions indicator and review of determinations The text then ponders on compounds of trivalent copper and potassium hexacyanoferrate The book ponders on hypohalites hypochlorite and hypobromite chloramine T and bromine as well as standard solutions indicator and review of determinations The publication also takes a look at iodine monochloride periodic acid and its salts lead IV acetate compounds of pentavalent vanadium and iron III salts The compounds of trivalent cobalt hydrogen peroxide chromium II salts tin II chloride sodium arsenite and compounds of monovalent copper are also elaborated The publication is a reliable reference for readers interested in newer redox titrants

Food Analysis Suzanne Nielsen, 2003-04-30 This book provides information on the techniques needed to analyze foods in laboratory experiments All topics covered include information on the basic principles procedures advantages limitations and applications This book is ideal for undergraduate courses in food analysis and is also an invaluable reference to professionals in the food industry General information is provided on regulations standards labeling sampling and data handling as background for chapters on specific methods to determine the chemical composition and characteristics of foods Large expanded sections on spectroscopy and chromatography are also included Other methods and instrumentation such as thermal analysis selective electrodes enzymes and immunoassays are covered from the perspective of their use in the chemical analysis of foods A helpful Instructor's Manual is available to adopting professors

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Encyclopedia of Environmental Science and Engineering James R. Pfafflin, Edward N. Ziegler, 1992 First Published in 1992 Routledge is an imprint of Taylor Francis an informa company

The Journal of Industrial and Engineering Chemistry, 1922 Measurement of Temperature, Salinity, and Velocity of Water Through Electrolytic Conductivity Measurements L. L. Higgins, 1962 A new technique is studied for the simultaneous independent and continuous measurement of the temperature salinity and velocity at a point in a flowing water solution by means of electrical measurements of the electrolytic conductivity and dielectric constant of the medium The instrumentation is capable of high sensitivity and rapid response limited only by the physical size of the probe and small sensors are relatively easy to construct in a simple and durable configuration The methods are applied to measurements in sea water and tap water

Journal of Industrial and Engineering Chemistry, 1922

Basic Nursing Leslie S Treas, Judith M Wilkinson, 2013-09-04 Thinking Doing Caring In every chapter you ll first explore the theoretical knowledge behind the concepts principles and rationales Then you ll study the practical knowledge involved in the processes and finally you ll learn the skills and procedures Student resources

available at DavisPlus davisplus fadavis com **Geochemical Exploration 1982** G.R. Parslow, 2013-10-22 Developments in Economic Geology Volume 17 Geochemical Exploration 1982 provides an outline of several significant areas of technical communications in relation to the mining industry This book discusses the role of governments universities and industries in the search for and development of the natural resources Organized into 56 chapters this volume begins with an overview of the significant role that technical communications play in everyday activities This text then examines the geochemical case histories for soil and lake sediment surveys Other chapters consider the chemistry of deep ground waters from throughout the Athabasca Basin This book discusses as well the uranium mineralization of the McClean Lake Area deposits which can be described as belonging to two different facies The final chapter deals with the application of factor analysis for the purpose of identifying areas potentially favorable for uranium deposits This book is a valuable resource for scientists and mineral engineers

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