# Chemistry 12 Worksheet 2-2 LeChatelier's Principle

	(KEY)	
( RE	10	4
	JOTAL = 10 marlo.	
<u>e</u>		1

← 1.	In order to decide what effect a change in total pressure will have on an equilibrium system
( )	with gases, what is the first thing you should do when given the balanced equation?
. 1 )	- Add up the moles of gas on each side of
	the equation
2	Predict which way the following equilibrium systems will shift when the total pressure is
	increased.(NOTE: Some may have no shift)
(2)	a). Name + Ose = 2Nove gas Answer no shift
( コノ	b). 2503(e) + 02(e) = 250 ve Answer right
$\smile$	c). 4NH3(D+502(D = 4NO(D)+6H2O(D Auswer 10ft
3.	Which way will the following equilibrium shift if the total pressure on the system is
	decreased?
2(I)	9 meles gas 10 meles gas to counternat the decrease in pressure
	Answer right
4	Explain why a flask filled with NO <sub>2(g)</sub> and N <sub>2</sub> O <sub>4(g)</sub> will get durker when heated. Use the equation: N <sub>2</sub> O <sub>4(g)</sub> + heat = 2NO <sub>2(g)</sub> Clock brown
	COUNTY OFFICE AND ADDRESS OF THE PROPERTY OF T
(1)	- heating will cause the equil to shift right.
( ソ	This will produce more NOw which is dark brown. Hence
	the colour wall get darker.
4	State Le Chatelier's Principle.
	*************************************
(1)	When a stress is applied to a system at equilibrium, the equil's will shift (left or ngit) so as to counteract
(ソ	11. Shirt (1847 or regict 50 as to Counterall
	the imposed stress

# **Prentice Hall Chemistry Chapter 7 Assessment Answers**

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Water-Insoluble Drug Formulation Ron Liu, 2018-03-12 Properties and Formulation From Theory to Real World Application Scientists have attributed more than 40 percent of the failures in new drug development to poor biopharmaceutical properties particularly water insolubility Issues surrounding water insolubility can postpone or completely derail important new drug development Even the much needed reformulation of currently marketed products can be significantly affected by these challenges More recently it was reported that the percentage increased to 90% for the candidates of new chemical entities in the discovery stage and 75% for compounds under development In the most comprehensive resource on the topic this third edition of Water Insoluble Drug Formulation brings together a distinguished team of experts to provide the scientific background and step by step guidance needed to deal with solubility issues in drug development Twenty three chapters systematically describe the detailed discussion on solubility theories solubility prediction models the aspects of preformulation biopharmaceutics pharmacokinetics regulatory and discovery support of water insoluble drugs to various techniques used in developing delivery systems for water insoluble drugs This book includes more

than 15 water insoluble drug delivery systems or technologies illustrated with case studies and featuring oral and parenteral applications Highlighting the most current information and data available this seminal volume reflects the significant progress that has been made in nearly all aspects of this field The aim of this book is to provide a handy reference for pharmaceutical scientists in the handling of formulation issues related to water insoluble drugs In addition this book may be useful to pharmacy and chemistry undergraduate students and pharmaceutical and biopharmaceutical graduate students to enhance their knowledge in the techniques of drug solubilization and dissolution enhancement Children's Books in Print, 2007, 2006 ASHRAE Transactions American Society of Heating, Refrigerating and Air-Conditioning Engineers, 1994

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