The Lewis dot symbol consists of the symbol for the element surrounded by dot(s). What does the dot or dots represent?

electron configuration

valence electrons

atomic number

atomic mass.

core electrons

Which of the following contains ionic bonding?

CO

SrF a

,641

OCI 2

HCL

How many electrons are in a single bond?

O

1 2

4

6

4. How many electrons are in a triple bond?

0

1

2

6

5. Select the strongest bond in the following group.

C-S

C-0

C = C

CIN

C-F

6. Which is the correct order for increasing bond strength? (1) CEC (2) C=C (3) C-C

(3) < (2) < (1)

(3) < (1) < (2)

(1) < (2) < (3)

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H Kauffman

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