

Chemistry - Review Chapters 4-5: 6

Directions: Answer each of the following questions. You need not use complete sentences.

1. By what property did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE; 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE; 18

5. State the number of valence electrons in an atom of:

a. sulfur	b. calcium	c. chlorine	d. arsenic
6	2	7	5

6. Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10	b. 18	c. 36	d. 90
Ne, NEON	Ar, ARGON	Kr, KRYPTON	Th, THORIUM

7. List, by number, both the period and group of each of these elements.

	<u>Symbol</u>	<u>Period</u>	<u>Group</u>
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

8. Which of the following pairs of elements belong to the same period?

a. Na and Cl	b. Na and Li	c. Na and Cu	d. Na and Ne
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9. Which of the following pairs of elements belong to the same group?

a. H and He	b. Li and Be	c. C and Pb	d. Ga and Ge
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10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

f ORBITALS

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AN Whitehead



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Table of Contents Prentice Hall Chemistry Guide Answer Key

1. Understanding the eBook Prentice Hall Chemistry Guide Answer Key
 - The Rise of Digital Reading Prentice Hall Chemistry Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Prentice Hall Chemistry Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Prentice Hall Chemistry Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Prentice Hall Chemistry Guide Answer Key
 - Personalized Recommendations
 - Prentice Hall Chemistry Guide Answer Key User Reviews and Ratings
 - Prentice Hall Chemistry Guide Answer Key and Bestseller Lists

5. Accessing Prentice Hall Chemistry Guide Answer Key Free and Paid eBooks
 - Prentice Hall Chemistry Guide Answer Key Public Domain eBooks
 - Prentice Hall Chemistry Guide Answer Key eBook Subscription Services
 - Prentice Hall Chemistry Guide Answer Key Budget-Friendly Options
6. Navigating Prentice Hall Chemistry Guide Answer Key eBook Formats
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 - Prentice Hall Chemistry Guide Answer Key Compatibility with Devices
 - Prentice Hall Chemistry Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Prentice Hall Chemistry Guide Answer Key
 - Highlighting and Note-Taking Prentice Hall Chemistry Guide Answer Key
 - Interactive Elements Prentice Hall Chemistry Guide Answer Key
8. Staying Engaged with Prentice Hall Chemistry Guide Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Prentice Hall Chemistry Guide Answer Key
9. Balancing eBooks and Physical Books Prentice Hall Chemistry Guide Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Prentice Hall Chemistry Guide Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Prentice Hall Chemistry Guide Answer Key
 - Setting Reading Goals Prentice Hall Chemistry Guide Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Prentice Hall Chemistry Guide Answer Key
 - Fact-Checking eBook Content of Prentice Hall Chemistry Guide Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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