

14

THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description next to the item.

Column I

1. The constant relating unit temperature change to the change in volume of a fixed mass of gas at constant pressure
2. The volume of a fixed mass of gas at constant pressure as the absolute temperature is changed
3. The ratio of the molar heat capacity at constant pressure to the molar heat capacity at constant volume
4. The constant relating unit change in temperature to the change in pressure of a fixed mass of gas at constant volume
5. A constant of state that is the product of the universal gas constant and the molar mass
6. The probability of gas molecules escaping the surface of the gas continuously with pressure
7. An ideal gas that obeys the ideal gas law at all temperatures and pressures
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. The amount of heat brought into the system by a process at constant volume
10. Heat at constant volume

Column II

- a. constant pressure
- b. ideal gas constant R
- c. volume
- d. compressibility
- e. Boyle's law
- f. Boyle's constant
- g. constant heat capacity process
- h. isobaric
- i. isochoric
- j. distribution coefficient

B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. Subliming the solvent of a solid-solvent system from a solid solvent, temperature constant, pressure constant is:
 - a. the sublimation process
 - b. the evaporation process
 - c. the condensation process
 - d. the deposition process

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Guillaume Favre



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