

Name: \_\_\_\_\_

# CLASSIFYING CHEMICAL REACTIONS

Classify the following chemical reactions as one of the following: synthesis, decomposition, single-replacement, double-replacement, or combustion.

|                                                                         |                                                                                 |
|-------------------------------------------------------------------------|---------------------------------------------------------------------------------|
| $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$              | $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$                            |
| $2\text{K} + 2\text{H}_2\text{O} \rightarrow \text{H}_2 + 2\text{KOH}$  | $\text{CaCO}_3 + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{CO}_3$ |
| $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$                  | $\text{Ca} + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$                |
| $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$                    | $\text{Pb(NO}_3)_2 + 2\text{HCl} \rightarrow \text{PbCl}_2 + 2\text{HNO}_3$     |
| $\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu(NO}_3)_2$ | $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$                             |
| $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$              | $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$       |
| $\text{Pb(NO}_3)_2 + 2\text{K} \rightarrow \text{Pb} + 2\text{KNO}_3$   | $\text{H}_2\text{CO}_3 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$            |
| $2\text{Fe} + \text{O}_2 \rightarrow 2\text{FeO}$                       | $\text{Zn} + 2\text{HCl} \rightarrow \text{H}_2 + \text{ZnCl}_2$                |
| $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$ | $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$        |

# Pre Lab Answers To Classifying Chemical Reactions

**Zhe Xu**



## **Pre Lab Answers To Classifying Chemical Reactions:**

## Decoding **Pre Lab Answers To Classifying Chemical Reactions**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Pre Lab Answers To Classifying Chemical Reactions**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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