

Naming Ionic Compounds Practice Worksheet

Name the following ionic compounds:

- 1) NH_4Cl Ammonium chloride
- 2) $\text{Fe}(\text{NO}_3)_3$ Iron (III) Nitrate
- 3) TiBr_3 Titanium (III) Bromide
- 4) Cu_3P Copper (I) Phosphide
- 5) SnSe_2 Tin (IV) Selenide
- 6) GaAs Gallium Arsenide
- 7) $\text{Pb}(\text{SO}_4)_2$ Lead (IV) Sulfate
- 8) $\text{Be}(\text{HCO}_3)_2$ Beryllium Hydrogen Carbonate
- 9) $\text{Mn}_2(\text{SO}_3)_3$ Manganese (II) Sulfite
- 10) $\text{Al}(\text{CN})_3$ Aluminum Cyanide

Write the formulas for the following compounds:

- 11) chromium (VI) phosphate $\text{Cr}(\text{PO}_4)_2$
- 12) vanadium (IV) carbonate $\text{V}(\text{CO}_3)_2$
- 13) tin (II) nitrite $\text{Sn}(\text{NO}_2)_2$
- 14) cobalt (III) oxide Co_2O_3
- 15) titanium (II) acetate $\text{Ti}(\text{C}_2\text{H}_3\text{O}_2)_2$
- 16) vanadium (V) sulfide V_2S_5
- 17) chromium (III) hydroxide $\text{Cr}(\text{OH})_3$
- 18) lithium iodide LiI
- 19) lead (II) nitride Pb_3N_2
- 20) silver bromide AgBr

Practice Ionic And Polyatomic Compounds Answers

Pamela Mills,Amina El-Ashmawy



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An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

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