



COVALENT BONDING

Practice Problems

Do your answers to these very practice problems.

SECTION 8.1 MOLECULAR COMPOUNDS

1. Classify each of the following compounds as ionic or covalent.

- a. NaCl ionic b. CH_4 covalent c. HCl covalent
d. KNO_3 ionic e. H_2O covalent

2. Write the formula for the following molecular compounds.

- a. carbon dioxide b. sulfur dioxide c. carbon monoxide
d. nitrogen e. sulfur f. phosphorus

3. Write the Lewis structure for each of the following molecular compounds.

a. H_2O b. CH_4 c. CO_2 d. H_2S e. NH_3 f. C_2H_6 g. C_2H_4 h. C_2H_2 i. $\text{C}_2\text{H}_4\text{Cl}_2$ j. $\text{C}_2\text{H}_4\text{Cl}_2$ k. $\text{C}_2\text{H}_4\text{Cl}_2$ l. $\text{C}_2\text{H}_4\text{Cl}_2$ m. $\text{C}_2\text{H}_4\text{Cl}_2$ n. $\text{C}_2\text{H}_4\text{Cl}_2$ o. $\text{C}_2\text{H}_4\text{Cl}_2$ p. $\text{C}_2\text{H}_4\text{Cl}_2$ q. $\text{C}_2\text{H}_4\text{Cl}_2$ r. $\text{C}_2\text{H}_4\text{Cl}_2$ s. $\text{C}_2\text{H}_4\text{Cl}_2$ t. $\text{C}_2\text{H}_4\text{Cl}_2$ u. $\text{C}_2\text{H}_4\text{Cl}_2$ v. $\text{C}_2\text{H}_4\text{Cl}_2$ w. $\text{C}_2\text{H}_4\text{Cl}_2$ x. $\text{C}_2\text{H}_4\text{Cl}_2$ y. $\text{C}_2\text{H}_4\text{Cl}_2$ z. $\text{C}_2\text{H}_4\text{Cl}_2$

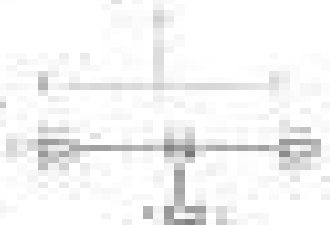
4. Draw the Lewis structure for each of the following molecular compounds. Be sure to show all lone pairs of electrons.

SECTION 8.2 THE NATURE OF COVALENT BONDING

1. Draw the Lewis structure for each of the following molecular compounds.



2. Draw the Lewis structure for each of the following molecular compounds.



3. Draw the Lewis structure for each of the following molecular compounds.



4. Draw the Lewis structure for each of the following molecular compounds.

5. Draw the Lewis structure for each of the following molecular compounds. Be sure to show all lone pairs of electrons.



SECTION 8.3 BONDING: THE ORBITAL

1. Draw the Lewis structure for each of the following molecular compounds. Be sure to show all lone pairs of electrons.

2. Draw the Lewis structure for each of the following molecular compounds. Be sure to show all lone pairs of electrons.

Practice Problems Chapter 8 Covalent Bonding Worksheet

Arshad Iqbal



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