

1. Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

Cu   **K**   Ni   **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in atomic radius as you go across a period is DECREASING. Therefore, the element on the far left (K) is the largest, and the element on the far right (Br) is the smallest.*

2. Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

Cu   **K**   Ni   **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in ionization energy as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest ionization energy, and the element on the far right (Br) has the highest ionization energy.*

3. Circle the element with the highest electronegativity and put a square around the element with the lowest electronegativity:

Cu   **K**   Ni   **Br**

- a. Explain why you made these choices: *All of the elements are in the same period. The trend in electronegativity as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest electronegativity, and the element on the far right (Br) has the highest electronegativity.*

4. For each of the following groups: Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

- 5.
- O   C   **Be**   **Ne**   Same Period
  - Na   Rb   **Fr**   **H**   Same Group
  - Pb**   **C**   Sn   Si   Same Group
  - Au   W   S   **Fr**   **Ne**   Zn   Challenge

6. For each of the following groups: Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

- O   C   **Be**   **Ne**   Same Period
- Na   Rb   **Fr**   **H**   Same Group
- Pb**   **C**   Sn   Si   Same Group
- Au   W   S   **Fr**   **Ne**   Zn   Challenge

# Periodic Table Trends Test Answer Key

**Thomas Evangelist**



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## **Table of Contents Periodic Table Trends Test Answer Key**

1. Understanding the eBook Periodic Table Trends Test Answer Key
  - The Rise of Digital Reading Periodic Table Trends Test Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Trends Test Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Trends Test Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Trends Test Answer Key
  - Personalized Recommendations
  - Periodic Table Trends Test Answer Key User Reviews and Ratings
  - Periodic Table Trends Test Answer Key and Bestseller Lists
5. Accessing Periodic Table Trends Test Answer Key Free and Paid eBooks
  - Periodic Table Trends Test Answer Key Public Domain eBooks
  - Periodic Table Trends Test Answer Key eBook Subscription Services
  - Periodic Table Trends Test Answer Key Budget-Friendly Options
6. Navigating Periodic Table Trends Test Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table Trends Test Answer Key Compatibility with Devices
  - Periodic Table Trends Test Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table Trends Test Answer Key
  - Highlighting and Note-Taking Periodic Table Trends Test Answer Key
  - Interactive Elements Periodic Table Trends Test Answer Key
8. Staying Engaged with Periodic Table Trends Test Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Periodic Table Trends Test Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Trends Test Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Trends Test Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Trends Test Answer Key
  - Setting Reading Goals Periodic Table Trends Test Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Trends Test Answer Key
  - Fact-Checking eBook Content of Periodic Table Trends Test Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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