STUDY GUIDE FOR CONTENT MASTERY

The Periodic Table and Periodic Law

Section 6.1 Development of the Modern Periodic Table

In your textbook, reads about the history of the periodic table's development.

Use each of the terms below just once to complete the passage.

octaves atomic mass atomic number nine elements properties Henry Moseley eight protons periodic law Dmitri Mendeleev accepted

1	2	3	4	5	6	7
H.	U	G	Bo	C	N	0
8	9	10	11	12	13	14
7	No.	Mg	Al	54	P	5

The first periodic table is mostly credited to (5) _______. In his table, the elements were arranged according to increasing (6) _______. One important result of this table was that the existence and properties of undiscovered

The element in the modern periodic table are arranged according to increasing

(8) ______, as a result of the work of (9) ______. This arrangement is based on number of (10) ______ in the nucleus of an atom of the element. The modern form of the periodic table results in the

(11)______, which states that when elements are arranged according to increasing atomic number, there is a periodic repetition of their chemical and physical

(12)______

(7) could be predicted.

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Periodic Table Teaching Transparency Answers

Cindy Blobaum

Periodic Table Teaching Transparency Answers:

Periodic Table Teaching Transparency Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Periodic Table Teaching Transparency Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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