

# Periodic Trends Worksheet

Using a few minutes before today's activity, complete the following questions. Please post your answers on the board when you are done.

Additional help on the periodic trends can be found on this video: <https://www.youtube.com/watch?v=0D8Wp-2GgY0>

1. What does electronegativity measure and how does it change across the periodic table (increased or decreased)? Explain, reference, substantiate, and justify.
2. Electronegativity increases from left to right across the periodic table. Explain.
3. Electronegativity increases from left to right across the periodic table. Explain your answer:
  - a. Fluorine is F
  - b. It is wrong
  - c. Cl is Cl
4. Electronegativity increases from left to right across the periodic table. Explain your answer:
  - a. It is F
  - b. Carbon is C
  - c. Li is Li
5. Electronegativity increases from left to right across the periodic table. Explain your answer:
  - a. Cl is Cl
  - b. Fluorine is F
  - c. It is Br
6. Electronegativity increases from left to right across the periodic table. Explain your answer:
  - a. Fluorine is F
  - b. Fluorine is F
  - c. It is Br
7. Electronegativity increases from left to right across the periodic table. Explain your answer:
  - a. It is F
  - b. Cl is Cl
  - c. Fluorine is F
8. Chlorine is a halogen. Explain all trends of the periodic table. Explain the trend of electronegativity.

# Periodic Trends Questions And Answers

**Ying Liu**



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education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner     *CCEA AS Unit 1 Chemistry Student Guide: Basic concepts in Physical and Inorganic Chemistry* Alyn G. McFarland, 2016-07-18 Reinforce students understanding throughout their course clear topic summaries with sample questions and answers will improve exam technique to achieve higher grades Written by examiners and teachers Student Guides Help students identify what they need to know with a concise summary of the topics examined in the AS and A level specification Consolidate understanding with exam tips and knowledge check questions Provide opportunities to improve exam technique with sample graded answers to exam style questions Develop independent learning and research skills Provide the content for generating individual revision notes     **Princeton Review AP Chemistry Prep, 25th Edition** The Princeton Review, 2023-08-01 Make sure you re studying with the most up to date prep materials Look for the newest edition of this title The Princeton Review AP Chemistry Premium Prep 26th Edition ISBN 9780593517604 on sale August 2024 Publisher s Note Products purchased from third party sellers are not guaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original product     **Chemistry in a Month** John Wilkinson, 2003



## **Periodic Trends Questions And Answers** Book Review: Unveiling the Power of Words

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