

Question 6 :
Use the below periodic table to answer the following questions :

Periodic Table of the Elements

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
A	B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R
S	T	U	V	W	X	Y	Z	AA	AB	AC	AD	AE	AF	AG	AH	AI	AJ
AK	AL	AM	AN	AO	AP	AQ	AR	AS	AT	AU	AV	AW	AX	AY	AZ	BA	BB
BC	BD	BE	BF	BG	BH	BI	BJ	BL	BM	BN	BO	BP	BQ	BR	BS	BT	BU
BV	BW	BX	BY	BZ	CA	CB	CC	CD	CE	CF	CG	CH	CI	CJ	CK	CL	CM
CN	CO	CP	CQ	CR	CS	CT	CU	CV	CW	CX	CY	CZ	DA	DB	DC	DD	DE
DF	DG	DH	DI	DJ	DK	DL	DM	DN	DO	DP	DQ	DR	DS	DT	DU	DV	DW
DX	DY	DX	DY	DX	DY	DX	DY	DX	DY	DX	DY	DX	DY	DX	DY	DX	DY
EA	EB	EC	ED	EE	EF	EG	EH	EI	EJ	EK	EL	EM	EN	EO	EP	EQ	ER
ES	ET	EU	EV	EW	EX	EY	EZ	FA	FB	FC	FD	FE	FF	FG	FH	FI	FJ
FK	FL	FM	FN	FO	FP	FQ	FR	FS	FT	FU	FV	FW	FX	FY	FZ	GA	GB
GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP	GQ	GR	GS	GT
GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM
GN	GO	GP	GQ	GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF
GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP	GQ	GR	GS	GT	GU	GV	GW	GX
GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP	GQ
GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ
GL	GM	GN	GO	GP	GQ	GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD
GN	GO	GP	GQ	GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF
GO	GP	GQ	GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG
GP	GQ	GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH
GQ	GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI
GR	GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ
GS	GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK
GT	GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL
GU	GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM
GV	GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN
GW	GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO
GX	GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP
GY	GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP	GQ
GA	GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP	GQ	GR
GB	GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP	GQ	GR	GS
GC	GD	GE	GF	GG	GH	GI	GJ	GK	GL	GM	GN	GO	GP	GQ	GR	GS	GT
GD	GE	GF	GG	GH	GI												

- Which element is located in group 2A, period 5?
- What is the period and the group of element (F) ?
- Which element has the highest Electronegativity ?
- Arrange (B , E , G) atoms in order of decreasing atomic radius.
- Arrange (B^{+2} , C^{+3} , D^{-1}) ions in order of decreasing atomic radius.
- Write the electron configuration of element (D) and its anion (D^{-}).
- Write the reaction of element (A) with water .

Periodic Table Test Questions Answers

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